



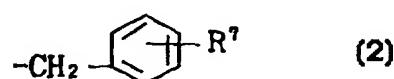
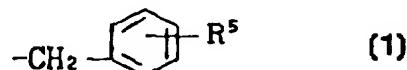
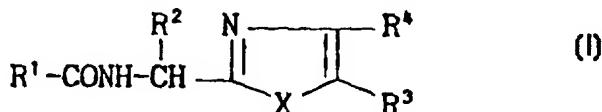
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(54) Title: NEW INDOLYL AND BENZOFURANYL CARBOXAMIDES AS INHIBITORS OF NITRIC OXIDE PRODUCTION

(57) Abstract

A compound of formula (I) wherein R¹ is indolyl or benzofuranyl; R² is hydrogen, lower alkylthio(lower)alkyl or a group of formula (1) in which R⁵ is hydrogen, lower alkoxy or halogen; R³ is hydrogen, quinolyl or phenyl which may have a suitable substituent selected from the group consisting of lower alkyl, lower alkoxy, lower alkylthio and halogen; R⁴ is hydrogen or optionally esterified carboxy; and X is S or NR⁶ in which R⁶ is hydrogen, lower alkyl or a group of formula (2) in which R⁷ is lower alkyl or lower alkoxy, and a pharmaceutically acceptable salt thereof, which possess a strong inhibitory activity on the production of nitric oxide (NO), and are useful for prevention and/or treatment of NO-mediated diseases such as adult respiratory distress syndrome, cardiovascular ischemia, myocarditis, heart failure, and the like.



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DESCRIPTION

NEW INDOLYL AND BENZOFURANYL CARBOXAMIDES AS INHIBITORS OF NITRIC OXIDE PRODUCTION

TECHNICAL FIELD

This invention relates to new amide compounds and pharmaceutically acceptable salts thereof which are useful as medicament.

BACKGROUND ART

Some peptide compounds have been known as described, for example, in EP 0 394 989 A2.

DISCLOSURE OF INVENTION

This invention relates to new amide compounds.

One object of this invention is to provide the new and useful amide compounds and pharmaceutically acceptable salts thereof which possess a strong inhibitory activity on the production of nitric oxide (NO).

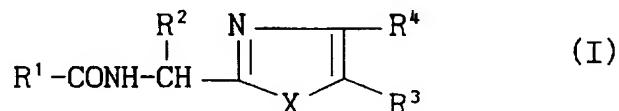
Another object of this invention is to provide a process for the preparation of the amide compounds and salts thereof.

A further object of this invention is to provide a pharmaceutical composition comprising said amide compound or a pharmaceutically acceptable salt thereof.

Still further object of this invention is to provide a use of said amide compounds or pharmaceutically acceptable salts thereof as a medicament for prophylactic and therapeutic treatment of NO-mediated diseases such as adult respiratory distress syndrome, cardiovascular ischemia, myocarditis, heart failure, synovitis, shock (e.g., septic shock, etc.), diabetes (e.g., insulin-dependent diabetes mellitus, etc.), diabetic nephropathy, diabetic retinopathy, diabetic neuropathy, glomerulonephritis, peptic ulcer, inflammatory bowel disease (e.g., ulcerative colitis, chronic colitis, etc.), cerebral

infarction, cerebral ischemia, cerebral hemorrhage, migraine, rheumatoid arthritis, gout, neuritis, postherpetic neuralgia, osteoarthritis, osteoporosis, systemic lupus erythematosis, rejection by organ transplantation, asthma, metastasis, Alzheimer's disease, arthritis, CNS disorders, and the like in human being and animals.

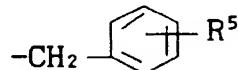
The object amide compounds of the present invention are novel and can be represented by the following general formula (I) :



wherein

R^1 is indolyl or benzofuranyl;

R^2 is hydrogen, lower alkylthio(lower)alkyl or a group of the formula:



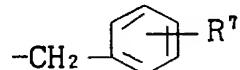
in which R^5 is hydrogen, lower alkoxy or halogen;

R^3 is hydrogen, quinolyl or phenyl which may have a suitable substituent selected from the group consisting of lower alkyl, lower alkoxy, lower alkylthio and halogen;

R^4 is hydrogen or optionally esterified carboxy; and

X is S or NR^6

in which R^6 is hydrogen, lower alkyl or a group of the formula:



in which R^7 is lower alkyl or lower alkoxy.

Suitable pharmaceutically acceptable salts of the object compound (I) are conventional non-toxic salts and include, for example, a salt with a base or an acid addition salt such as a salt with an inorganic

base, for example, an alkali metal salt (e.g., sodium salt, potassium salt, etc.), an alkaline earth metal salt (e.g., calcium salt, magnesium salt, etc.), an ammonium salt; a salt with an organic base, for example, an organic amine salt (e.g., triethylamine salt, pyridine salt, picoline salt, ethanolamine salt, triethanolamine salt, dicyclohexylamine salt, N,N'-dibenzylethylenediamine salt, etc.); an inorganic acid addition salt (e.g., hydrochloride, hydrobromide, sulfate, phosphate, etc.); an organic carboxylic or sulfonic acid addition salt (e.g., formate, acetate, trifluoroacetate, maleate, tartrate, citrate, fumarate, methanesulfonate, benzenesulfonate, toluenesulfonate, etc.); and a salt with a basic or acidic amino acid (e.g., arginine, aspartic acid, glutamic acid, etc.).

In the above and subsequent descriptions of the present specification, suitable examples and illustration of the various definitions which the present invention intends to include within the scope thereof are explained in detail as follows.

The term "lower" is used to intend a group having 1 to 6, preferably 1 to 4, carbon atom(s), unless otherwise provided.

Suitable "lower alkyl" and "lower alkyl moiety" in the terms "lower alkylthio" and "lower alkylthio(lower)alkyl" include straight or branched one having 1 to 6 carbon atom(s), such as methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, tert-butyl, pentyl, tert-pentyl and hexyl, and in which more preferred one is C₁-C₄ alkyl.

Suitable "lower alkoxy" includes, for example, methoxy, ethoxy, propoxy, isopropoxy, butoxy, isobutoxy, tert-butoxy, pentyloxy, tert-pentyloxy and hexyloxy, and in which more preferred one is C₁-C₄ alkoxy.

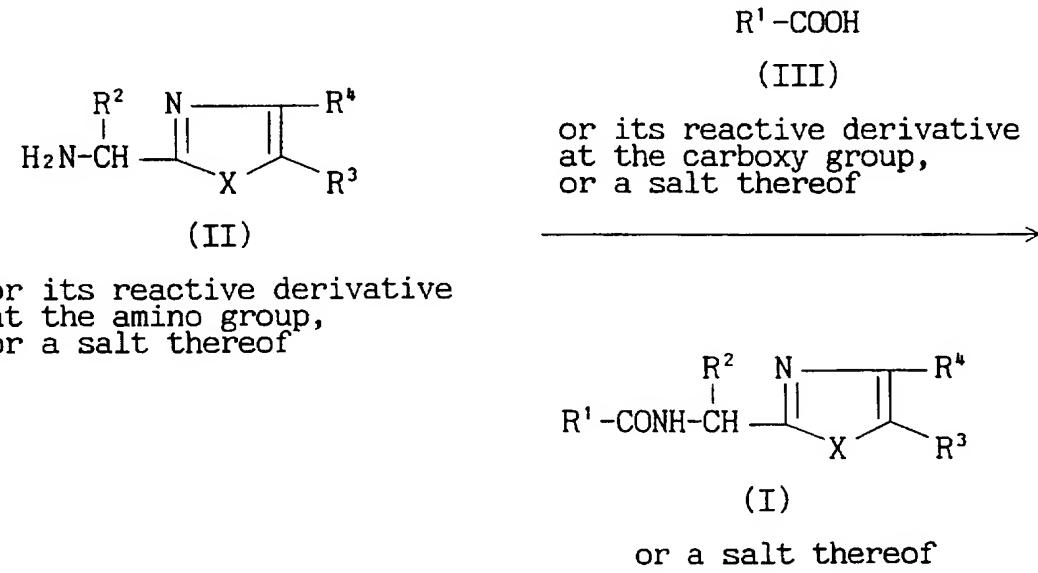
Suitable "halogen" includes, for example, fluorine, bromine, chlorine and iodine.

"Optionally esterified carboxy" includes carboxy and esterified carboxy. Suitable examples of said ester include lower alkyl ester

(e.g., methyl ester, ethyl ester, propyl ester, isopropyl ester, butyl ester, isobutyl ester, tert-butyl ester, pentyl ester, tert-pentyl ester, hexyl ester, etc.); lower alkenyl ester (e.g., vinyl ester, allyl ester, etc.); lower alkynyl ester (e.g., ethynyl ester, propynyl ester, etc.); lower alkoxy(lower)alkyl ester (e.g., methoxymethyl ester, ethoxymethyl ester, isopropoxymethyl ester, 1-methoxyethyl ester, 1-ethoxyethyl ester, etc.); mono(or di or tri)-aryl(lower)alkyl ester, for example, mono(or di or tri)phenyl(lower)-alkyl ester which may have one or more suitable substituent(s) [e.g., benzyl ester, 4-methoxybenzyl ester, 4-nitrobenzyl ester, phenethyl ester, trityl ester, benzhydryl ester, bis(methoxyphenyl)methyl ester, 3,4-dimethoxybenzyl ester, 4-hydroxy-3,5-di-tert-butylbenzyl ester, etc.]; and aryl ester which may have one or more suitable substituent(s) such as substituted or unsubstituted phenyl ester (e.g., phenyl ester, tolyl ester, tert-butylphenyl ester, xylyl ester, mesityl ester, cumenyl ester, 4-chlorophenyl ester, 4-methoxyphenyl ester, etc.).

The object compound (I) of the present invention can be prepared by the following process.

Process (1)



wherein R¹, R², R³, R⁴ and X are each as defined above.

The starting compounds can be prepared by the method of Preparation mentioned below or by a process known in the art for preparing structually analogous compounds thereto.

The process for preparing the object compound is explained in detail in the following.

Process (1)

The compound (I) or a salt thereof can be prepared by reacting the compound (II) or its reactive derivative at the amino group, or a salt thereof with the compound (III) or its reactive derivative at the carboxy group, or a salt thereof.

Suitable reactive derivative of the compound (II) includes Schiff's base type imino or its tautomeric enamine type isomer formed by the reaction of the compound (II) with a carbonyl compound such as aldehyde, ketone or the like; a silyl derivative formed by the reaction of the compound (II) with a silyl compound such as N,O-bis(trimethylsilyl)acetamide, N-trimethylsilylacetamide or the like; a derivative formed by the reaction of the compound (II) with phosphorus trichloride or phosgene.

Suitable reactive derivative of the compound (III) includes an acid halide, an acid anhydride and an activated ester. The suitable example may be an acid chloride; an acid azide; a mixed acid anhydride with an acid such as substituted phosphoric acid (e.g., dialkylphosphoric acid, phenylphosphoric acid, diphenylphosphoric acid, dibenzylphosphoric acid, halogenated phosphoric acid, etc.), dialkylphosphorous acid, sulfurous acid, thiosulfuric acid, alkanesulfonic acid (e.g., methanesulfonic acid, ethanesulfonic acid, etc.), sulfuric acid, alkylcarbonic acid, aliphatic carboxylic acid

(e.g., pivalic acid, pentanoic acid, isopentanoic acid, 2-ethylbutyric acid, trichloroacetic acid, etc.); aromatic carboxylic acid (e.g., benzoic acid, etc.); a symmetrical acid anhydride; an activated amide with imidazole, 4-substituted imidazole, dimethylpyrazole, triazole or tetrazole; an activated ester (e.g., cyanomethyl ester, methoxymethyl ester, dimethyliminomethyl $[(\text{CH}_3)_2\text{N}^+=\text{CH}-]$ ester, vinyl ester, propargyl ester, p-nitrophenyl ester, 2,4-dinitrophenyl ester, trichlorophenyl ester, pentachlorophenyl ester, mesylphenyl ester, phenylazophenyl ester, phenyl thioester, p-nitrophenyl thioester, p-cresyl thioester, carboxymethyl thioester, pyranyl ester, pyridyl ester, piperidyl ester, 8-quinolyl thioester, etc.); or an ester with an N-hydroxy compound (e.g., N,N-dimethylhydroxylamine, 1-hydroxy-2-(1H)-pyridone, N-hydroxysuccinimide, N-hydroxybenzotriazole, N-hydroxyphthalimide, 1-hydroxy-6-chloro-1H-benzotriazole, etc.).

These reactive derivatives can optionally be selected from them according to the kind of the compound (III) to be used.

The reaction is usually carried out in a conventional solvent such as water, acetone, dioxane, acetonitrile, chloroform, methylene chloride, ethylene chloride, tetrahydrofuran, ethyl acetate, N,N'-dimethylformamide, pyridine or any other organic solvents which do not adversely affect the reaction, or the mixture thereof.

When the compound (III) is used in free acid form or its salt form in the reaction, the reaction is preferably carried out in the presence of a conventional condensing agent such as N,N'-dicyclohexylcarbodiimide; N-cyclohexyl-N'-morpholinoethylcarbodiimide; N-cyclohexyl-N'-(4-diethylaminocyclohexyl)carbodiimide; N,N'-diisopropylcarbodiimide; N-ethyl-N'-(3-dimethylaminopropyl)-carbodiimide; N,N-carbonyl-bis-(2-methylimidazole); pentamethylene-ketene-N-cyclohexylimine; diphenylketene-N-cyclohexylimine; ethoxyacetylene; 1-alkoxy-1-chloroethylene; trialkyl phosphite; isopropyl polyphosphate; phosphorus oxychloride (phosphoryl chloride); phosphorus trichloride; thionyl chloride; oxalyl chloride;

triphenylphosphine; 2-ethyl-7-hydroxybenzisoxazolium salt; 2-ethyl-5-(*m*-sulfophenyl)isoxazolium hydroxide intramolecular salt; 1-(*p*-chlorobenzenesulfonyloxy)-6-chloro-1*H*-benzotriazole; so-called Vilsmeier reagent prepared by the reaction of N,N-dimethylformamide with thionyl chloride, phosgene, phosphorus oxychloride, etc.; or the like.

The reaction may also be carried out in the presence of an organic or inorganic base such as an alkali metal bicarbonate, tri(lower)alkylamine, pyridine, N-(lower)alkylmorpholine, N,N-di(lower)alkylbenzylamine, or the like.

The reaction temperature is not critical, and the reaction is usually carried out under cooling to heating.

Suitable salts of the starting compounds and their reactive derivatives in Process (1) can be referred to the ones as exemplified for the compound (I).

The compounds obtained by the above process can be isolated and purified by a conventional method such as pulverization, recrystallization, column chromatography, reprecipitation, or the like.

It is to be noted that the compound (I) and the other compounds may include one or more stereoisomer(s) such as optical isomer(s) and geometrical isomer(s) due to asymmetric carbon atom(s) and double bond(s), and all of such isomers and mixtures thereof are included within the scope of this invention.

The object compounds (I) and pharmaceutically acceptable salts thereof include solvates [e.g., enclosure compounds (e.g., hydrate, etc.)].

The object compounds (I) and pharmaceutically acceptable salts thereof possess a strong inhibitory activity on the production of nitric oxide (NO).

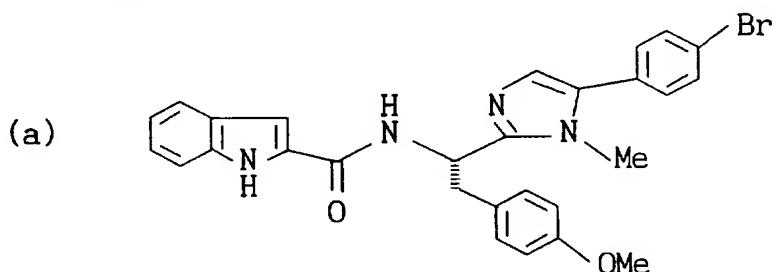
Accordingly, the object compounds (I) and pharmaceutically

acceptable salts thereof are expected to possess a nitric oxide synthase (NOS)-inhibitory activity or a NOS-production inhibitory activity.

Accordingly, they are useful for prevention and/or treatment of NO-mediated diseases such as adult respiratory distress syndrome, cardiovascular ischemia, myocarditis, heart failure, synovitis, shock (e.g., septic shock, etc.), diabetes (e.g., insulin-dependent diabetes mellitus, etc.), diabetic nephropathy, diabetic retinopathy, diabetic neuropathy, glomerulonephritis, peptic ulcer, inflammatory bowel disease (e.g., ulcerative colitis, chronic colitis, etc.), cerebral infarction, cerebral ischemia, cerebral hemorrhage, migraine, rheumatoid arthritis, gout, neuritis, postherpetic neuralgia, osteoarthritis, osteoporosis, systemic lupus erythematosis, rejection by organ transplantation, asthma, metastasis, Alzheimer's disease, arthritis, CNS disorders, and the like in human being and animals.

In order to illustrate the usefulness of the object compound (I), the pharmacological test result of the representative compound of the compound (I) is shown in the following.

Test Compound :



Test : Assay for inhibitory activity on the production
of nitric oxide

The murine macrophage cell line RAW264.7 (American Type Culture Collection, No. TIB71) was used in this study. RAW264.7 cells were grown on F75 plastic culture flasks at 37°C, 5% in Dulbecco's

modified Eagle's medium (DMEM) supplemented with L-glutamine, penicillin, streptomycin and 10% heat-inactivated fetal bovine serum. They were removed from culture flasks by rubber cell scraper and were centrifuged and resuspended in DMEM without phenol red. They were plated in 96-well microtiter plates (10^5 cells per well) and allowed to adhere over 2 hours. The test samples were added and the cells were preincubated for 1 hour. Thereafter the cells were activated with both of lipopolysaccharide (LPS) ($1 \mu\text{g}/\text{ml}$) and interferon γ (INF γ) ($3 \text{ u}/\text{ml}$) for 18-24 hours. An equal volume of Griess reagent (1% sulfanilamide/0.1% N-naphthylethylenediamine dihydrochloride/2.5% H₃PO₄) was added and the cells were incubated at room temperature for 10 minutes. The absorbance was read at 570 nm using microplate reader and NO₂⁻ was measured using NaNO₂ as a standard.

Test result :

Test compound (10^{-5}M)	Inhibition (%)
(a)	100

For therapeutic administration, the object compound (I) of the present invention and pharmaceutically acceptable salts thereof are used in the form of a conventional pharmaceutical preparation in admixture with a conventional pharmaceutically acceptable carrier such as an organic or inorganic solid or liquid excipient which is suitable for oral, parenteral or external administration. The pharmaceutical preparation may be compounded in a solid form such as granule, capsule, tablet, dragee or suppository, or in a liquid form such as solution, suspension or emulsion for injection, intravenous drip, ingestion, eye drop, etc. If needed, there may be included in the above preparation auxiliary substance such as stabilizing agent, wetting or emulsifying agent, buffer or any other commonly used additives.

The effective ingredient may usually be administered in a unit dose of 0.001 mg/kg to 500 mg/kg, preferably 0.01 mg/kg to 10 mg/kg, 1 to 4 times a day. However, the above dosage may be increased or decreased according to age, body weight and conditions of the patient or administering method.

The following Preparations and Examples are given for the purpose of illustrating the present invention in detail.

In the following Examples and Preparations, there are employed the other abbreviations in addition to the abbreviations adopted by the IUPAC-IUB (Commission on Biological Nomenclature).

The abbreviations used are as follows.

Boc : t-butoxycarbonyl

Et : ethyl

Me : methyl

Ph : phenyl

Ts : p-toluenesulfonyl

The starting compounds used and the object compounds obtained in the following Preparations and Examples are given in the Tables as below, in which the formulae of the starting compounds are in the upper and the formulae of the object compounds are in the lower, respectively.

Table

Preparation No.	Formula
1	
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3	
3	

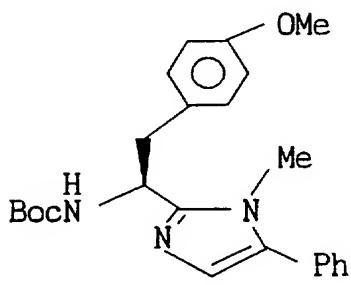
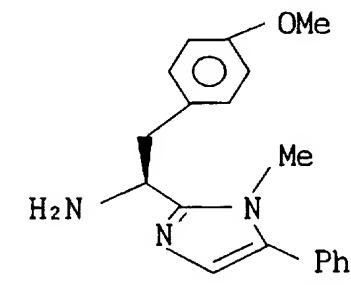
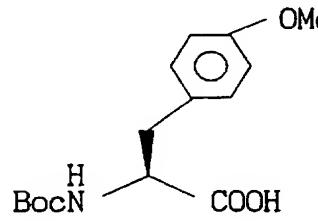
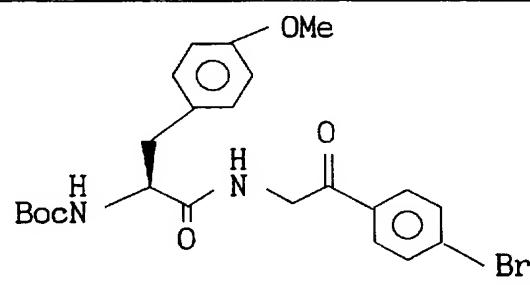
Table

Preparation No.	Formula
4	
5	
6	

Table

Preparation No.	Formula
7	<p>Structure of compound 7: A chiral center with a phenyl ring substituted at the para position with an OMe group. Attached to the chiral center is a tert-butyl carbamate group (BocN) and a carboxylic acid group (COOH).</p>
	<p>Structure of compound 8: Similar to compound 7, but the carboxylic acid group is replaced by a phenylacetyl amide group (-CONHCH₂CH₂Ph).</p>
8	<p>Structure of compound 9: Similar to compound 8, but the phenylacetyl amide group is replaced by a 2-methylimidazole ring attached to the phenyl ring via its 4-position.</p>

Table

Preparation No.	Formula
9	
	
10	
	

Table

Preparation No.	Formula
11	<p>Chemical structure of compound 11: A chiral amine derivative. It features a central carbon atom bonded to a phenyl ring substituted with a methoxy group (-OMe), a hydrogen atom, an amino group (-NH2), and a BocN group (tert-butyloxycarbonyl). The nitrogen atom of the amino group is also bonded to a phenyl ring substituted with a bromine atom (-Br).</p>
12	<p>Chemical structure of compound 12: A chiral amine derivative. It features a central carbon atom bonded to a phenyl ring substituted with a methoxy group (-OMe), a hydrogen atom, an amino group (-NH2), and a BocN group (tert-butyloxycarbonyl). The nitrogen atom of the amino group is also bonded to a phenyl ring substituted with a bromine atom (-Br).</p>
	<p>Chemical structure of compound 13: A chiral amine derivative. It features a central carbon atom bonded to a phenyl ring substituted with a methoxy group (-OMe), a hydrogen atom, an amino group (-NH2), and a BocN group (tert-butyloxycarbonyl). The nitrogen atom of the amino group is also bonded to a phenyl ring substituted with a bromine atom (-Br).</p>

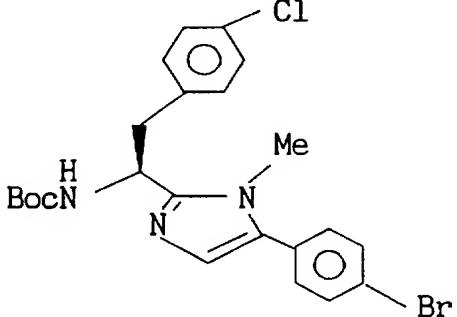
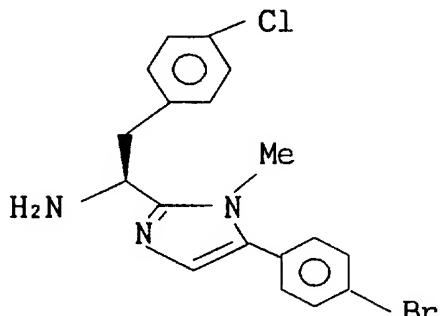
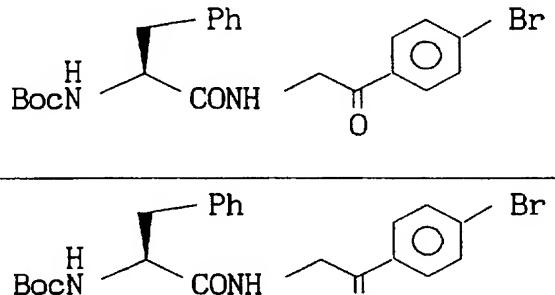
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Preparation No.	Formula
13	
14	
15	

Table

Preparation No.	Formula
16	
17	

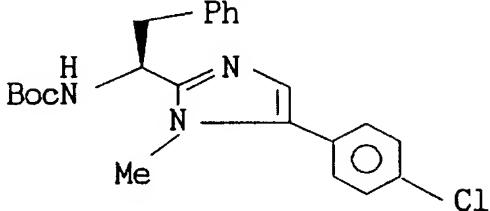
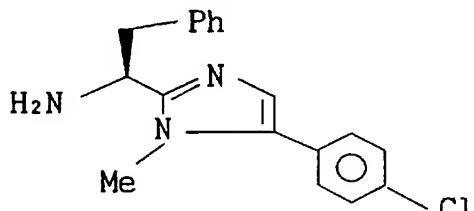
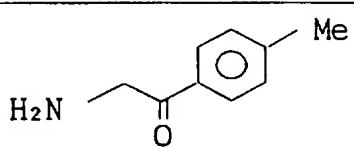
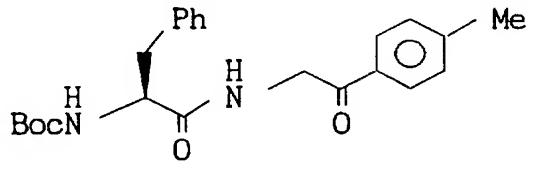
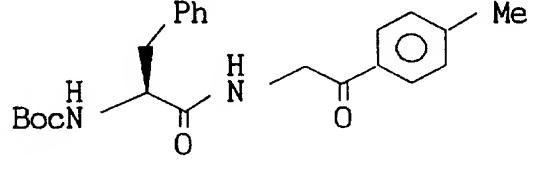
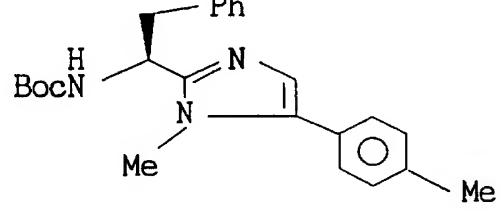
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Preparation No.	Formula
18	 <p>Chemical structure of compound 18: A pyrazine ring substituted at position 2 with a 4-chlorophenyl group, at position 4 with a 4-bromophenyl group, and at position 6 with a (S)-Boc-aminoethyl group.</p>
19	 <p>Chemical structure of compound 19: A pyrazine ring substituted at position 2 with a 4-chlorophenyl group, at position 4 with a 4-bromophenyl group, and at position 6 with a (S)-aminoethyl group.</p>
20	 <p>Chemical structure of compound 20: A pyrazine ring substituted at position 2 with a 4-chlorophenyl group, at position 4 with a 4-bromophenyl group, and at position 6 with a (S)-Boc-(1-phenylpropyl)aminoethyl group.</p>

Table

Preparation No.	Formula
21	
22	
23	

Table

Preparation No.	Formula
24	
	
25	
	
26	
	

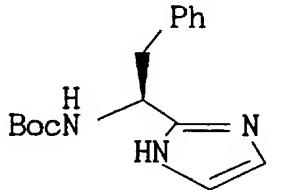
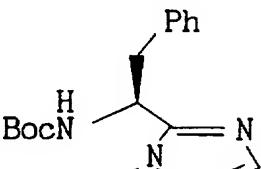
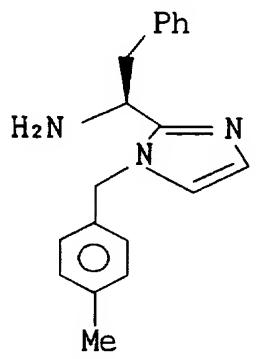
Table

Preparation No.	Formula
27	<p>Chemical structure of compound 27: A pyrazine ring substituted with a chiral phenyl group at position 2, a methyl group at position 4, and a 4-methylphenyl group at position 6.</p>
	<p>Chemical structure of compound 28: A pyrazine ring substituted with a chiral phenyl group at position 2, a methyl group at position 4, and a 4-chlorophenyl group at position 6.</p>
28	<p>Chemical structure of compound 29: A chiral phenyl group attached to a central carbon atom, which is also bonded to a BocN group, a COOH group, and a 4-chlorophenyl group.</p>
	<p>Chemical structure of compound 30: A chiral phenyl group attached to a central carbon atom, which is also bonded to a BocN group, an amide group (H-N-C(=O)-), and a 4-phenylbutyryl group.</p>

Table

Preparation No.	Formula
29	<p>Chemical structure of compound 29: A chiral amine derivative. It features a central carbon atom bonded to a phenyl ring substituted with a chlorine atom (Cl), a BocN group, a hydrogen atom (H), and an acetyl group (-CH₂COPh).</p>
30	<p>Chemical structure of compound 30: A chiral amine derivative. It features a central carbon atom bonded to a phenyl ring substituted with a chlorine atom (Cl), a BocN group, a hydrogen atom (H), and a 2-phenylpyrimidine ring substituted with a methyl group (Me).</p>
31	<p>Chemical structure of compound 31: A chiral aldehyde derivative. It features a central carbon atom bonded to a phenyl ring (Ph), a BocN group, a hydrogen atom (H), and an aldehyde group (-CHO).</p>
	<p>Chemical structure of compound 31: A chiral imine derivative. It features a central carbon atom bonded to a phenyl ring (Ph), a BocN group, a hydrogen atom (H), and a 2-phenylpyrimidine ring.</p>

Table

Preparation No.	Formula
32	 <p>Chemical structure of compound 32: A pyrazine ring substituted at position 2 with a (S)-Boc-aminoethyl group (Ph and H) and at position 4 with a 2-(4-methylphenyl)ethyl group.</p>
33	 <p>Chemical structure of compound 33: A pyrazine ring substituted at position 2 with a (S)-Boc-aminoethyl group (Ph and H) and at position 4 with a 2-(4-methylphenyl)ethyl group.</p>
	 <p>Chemical structure of compound 33: A pyrazine ring substituted at position 2 with an (S)-aminoethyl group (Ph and H₂N) and at position 4 with a 2-(4-methylphenyl)ethyl group.</p>

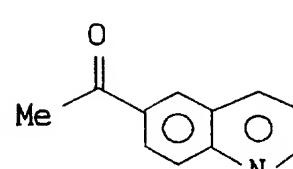
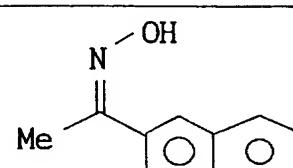
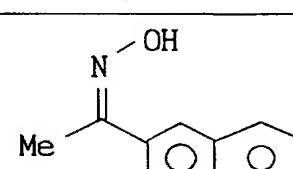
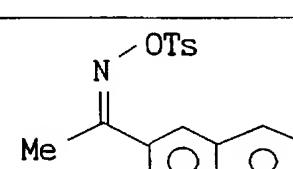
Table

Preparation No.	Formula
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Table

Preparation No.	Formula
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39	

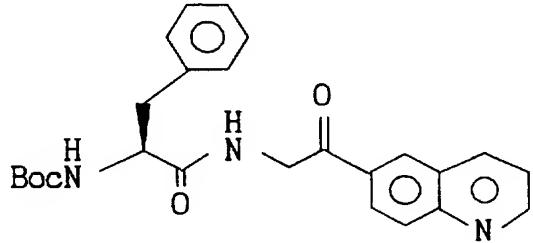
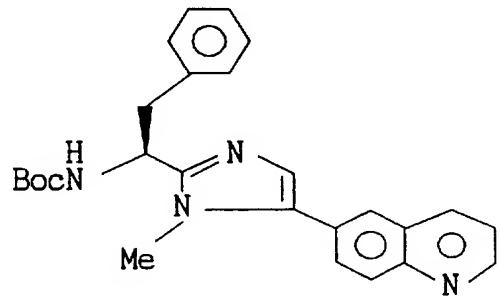
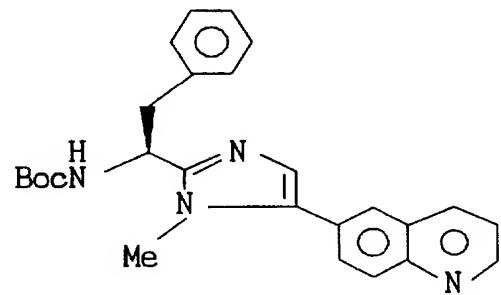
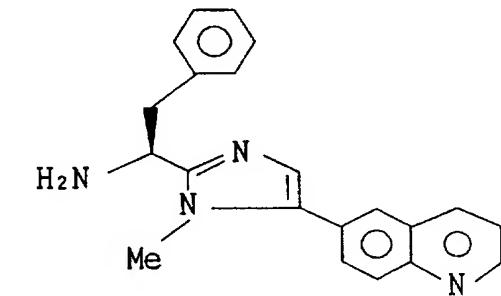
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Preparation No.	Formula
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41	
	

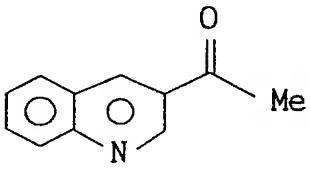
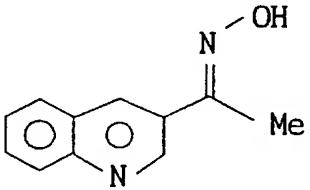
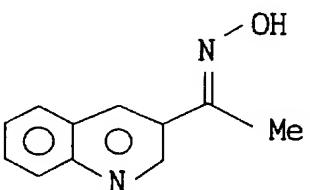
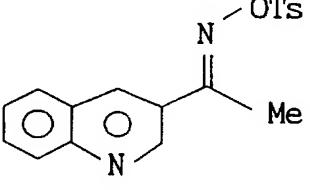
Table

Preparation No.	Formula
42	<p>The structure shows a 1H-isoindole ring system. At position 2, there is a carbonyl group (C=O) attached to a methyl group (Me) and a nitro group (-N+OTs). The nitrogen atom of the isoindole is also bonded to a methyl group.</p>
	<p>The structure shows a 1H-isoindole ring system. At position 2, there is a carbonyl group (C=O) attached to a 2-aminoethyl group (-CH2NH2) and a 2-hydroxyethyl group (-CH2OH). The nitrogen atom of the isoindole is bonded to a methyl group. The salt form is indicated as · 2HCl.</p>
43	<p>The structure shows a 1H-isoindole ring system. At position 2, there is a carbonyl group (C=O) attached to a 2-aminoethyl group (-CH2NH2) and a 2-hydroxyethyl group (-CH2OH). The nitrogen atom of the isoindole is bonded to a methyl group. The salt form is indicated as · 2HCl.</p>
	<p>The structure shows a 1H-isoindole ring system. At position 2, there is a carbonyl group (C=O) attached to a 2-aminoethyl group (-CH2NH2) and a 2-hydroxyethyl group (-CH2OH). The nitrogen atom of the isoindole is bonded to a methyl group. A BocN group is attached to the 2-aminoethyl side chain, and a phenyl group is attached to the 2-hydroxyethyl side chain via a chiral center indicated by a wedge bond.</p>

Table

Preparation No.	Formula
44	
	
45	
	

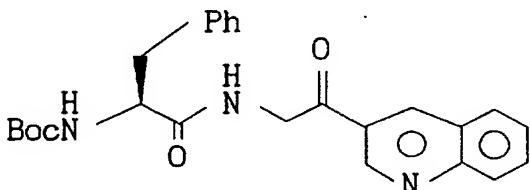
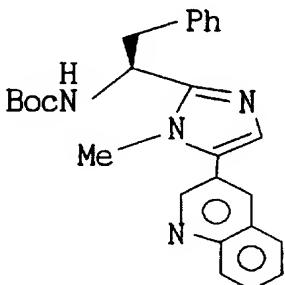
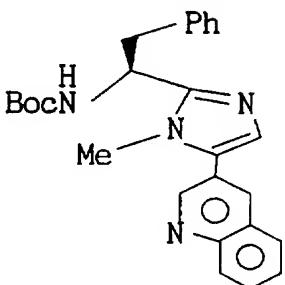
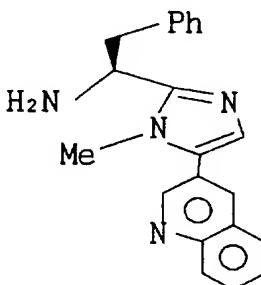
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Preparation No.	Formula
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Table

Preparation No.	Formula
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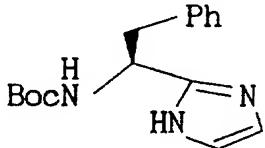
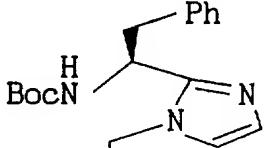
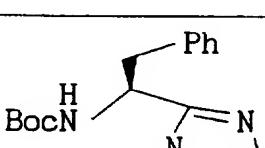
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Preparation No.	Formula
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51	
	

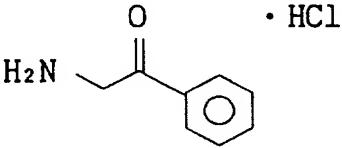
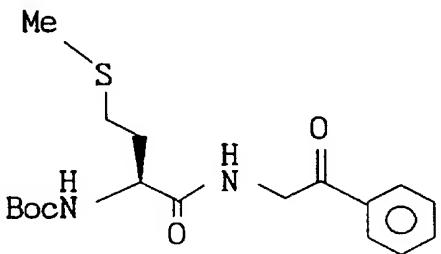
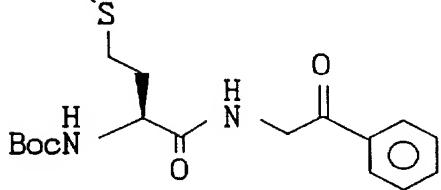
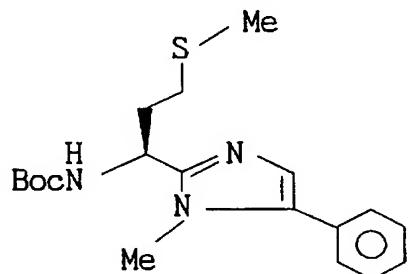
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Preparation No.	Formula
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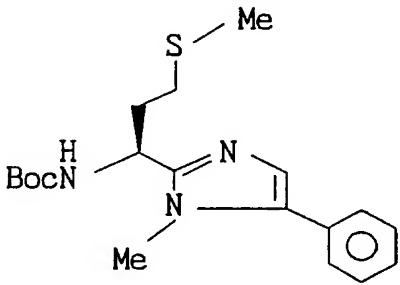
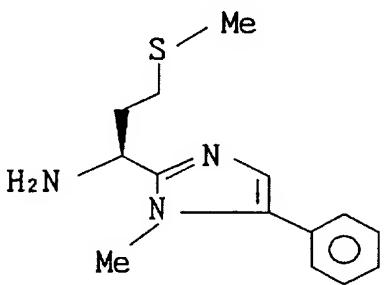
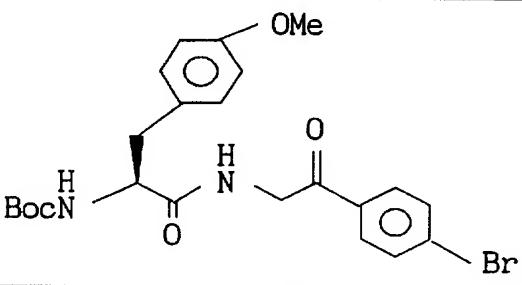
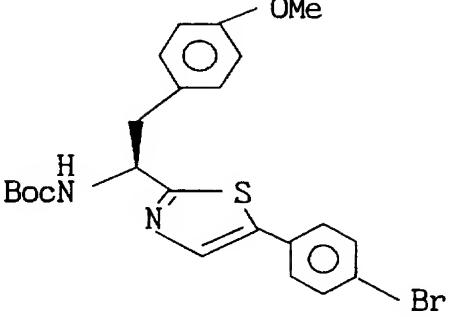
Table

Preparation No.	Formula
55	 <p>Chemical structure of compound 55: A pyrazine ring substituted at position 2 with a 2-((S)-Boc-1-phenylpropyl)amino group.</p>
56	 <p>Chemical structure of compound 56: A pyrazine ring substituted at position 2 with a 2-((S)-Boc-1-(4-methoxyphenyl)propyl)amino group.</p>
	 <p>Chemical structure of compound 57: A pyrazine ring substituted at position 2 with a 2-((S)-1-(4-methoxyphenyl)propyl)amino group.</p>

Table

Preparation No.	Formula
57	 <p style="text-align: right;">• HCl</p>
	
58	
	

Table

Preparation No.	Formula
59	 <p>Chemical structure of compound 59: A pyrazine ring substituted at position 2 with a 4-phenylbutyl group, at position 4 with a methyl group, and at position 6 with a (S)-2-methylthiethyl group.</p>
60	 <p>Chemical structure of compound 60: A pyrazine ring substituted at position 2 with a 4-phenylbutyl group, at position 4 with a methyl group, and at position 6 with an (R)-2-methylthiethyl group.</p>
60	 <p>Chemical structure of compound 60 derivative: A pyrazine ring substituted at position 2 with a 4-(4-methoxyphenyl)butyl group, at position 4 with a methyl group, and at position 6 with a 4-(4-bromophenyl)butyryl group.</p>
	 <p>Chemical structure of compound 60 derivative: A pyrazine ring substituted at position 2 with a 4-(4-methoxyphenyl)butyl group, at position 4 with a methyl group, and at position 6 with a 4-(4-bromophenyl)methylthio group.</p>

Table

Preparation No.	Formula
61	<p>Structure of compound 61: A chiral thiazole derivative. It features a central thiazole ring substituted at the 2-position with a 4-(4-bromophenoxy)butyl group and at the 4-position with a 4-methoxyphenyl group. The nitrogen atom is bonded to a chiral center with a BocN group and an H atom.</p>
62	<p>Structure of compound 62a: N-(4-ethoxyphenyl)-2-hydrazinylbutanamide. Structure of compound 62b: N-(4-ethoxyphenyl)-2-(BocN-1-phenylpropyl)acetamide.</p>
63	<p>Structure of compound 63a: N-(4-ethoxyphenyl)-2-(BocN-1-phenylpropyl)acetamide. Structure of compound 63b: 2-(BocN-1-phenylpropyl)-4-(4-ethoxyphenyl)-6-methylpyrimidine.</p>

Table

Preparation No.	Formula
64	<p>Structure of compound 64: A pyrazine ring with a phenyl group at position 2 and a methyl group at position 4. At position 6, there is a 4-(ethoxyphenyl)but-1-en-3-ynyl group. The ethoxyphenyl group has an ethoxy (-OEt) substituent.</p>
65	<p>Structure of compound 65: A pyrazine ring with a phenyl group at position 2 and a methyl group at position 4. At position 6, there is an ethyl group.</p>
66	<p>Structure of compound 66: A pyrazine ring with a phenyl group at position 2 and a methyl group at position 4. At position 6, there is an ethyl group.</p>

Table

Preparation No.	Formula
67	
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69	

Table

Preparation No.	Formula
70	
71	

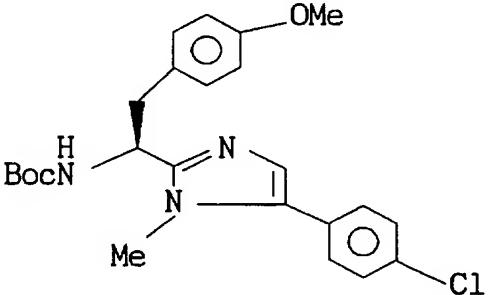
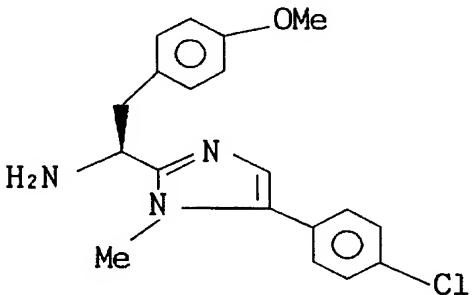
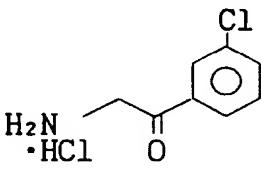
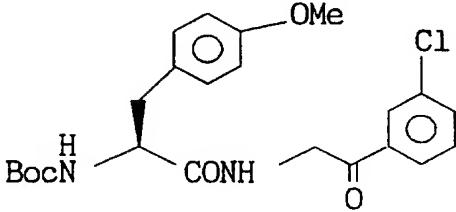
Table

Preparation No.	Formula
72	
73	

Table

Preparation No.	Formula
74	<p>$\text{H}_2\text{N} \cdot \text{HCl}$ Cl</p>
75	<p>BocN H CONH</p>
75	<p>BocN H CONH</p>

Table

Preparation No.	Formula
76	 <p>Structure of Preparation 76: A pyrazine ring substituted at position 2 with a 4-methoxyphenyl group, at position 4 with a 4-chlorophenyl group, and at position 6 with a (S)-Boc-1-hydroxypropan-2-yl group.</p>
	 <p>Structure of Preparation 76: A pyrazine ring substituted at position 2 with a 4-methoxyphenyl group, at position 4 with a 4-chlorophenyl group, and at position 6 with a (R)-1-amino-2-hydroxypropan-2-yl group.</p>
77	 <p>Structure of Preparation 77: (S)-1-(4-chlorophenyl)-2-hydrazinylpropane dihydrochloride.</p>
	 <p>Structure of Preparation 77: (S)-1-(4-chlorophenyl)-2-hydrazinylpropane dihydrochloride and its derivative where the hydrazinyl group is replaced by a CONH-Boc group.</p>

Table

Preparation No.	Formula
78	<p>Chemical structure of compound 78:</p> <pre> OMe C[C@H](c1ccc(O)cc1)N(C)C(=O)c2ccc(Cl)cc2 H2N </pre>
79	<p>Chemical structure of compound 79:</p> <pre> OMe C[C@H](c1ccc(O)cc1)c2nc(C)c(C=Cc3ccc(Cl)cc3)nc2Me H2N </pre>
	<p>Chemical structure of compound 79:</p> <pre> OMe C[C@H](c1ccc(O)cc1)c2nc(C)c(C=Cc3ccc(Cl)cc3)nc2Me H2N </pre>

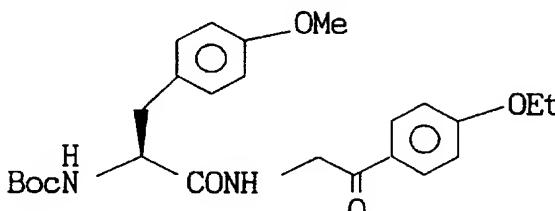
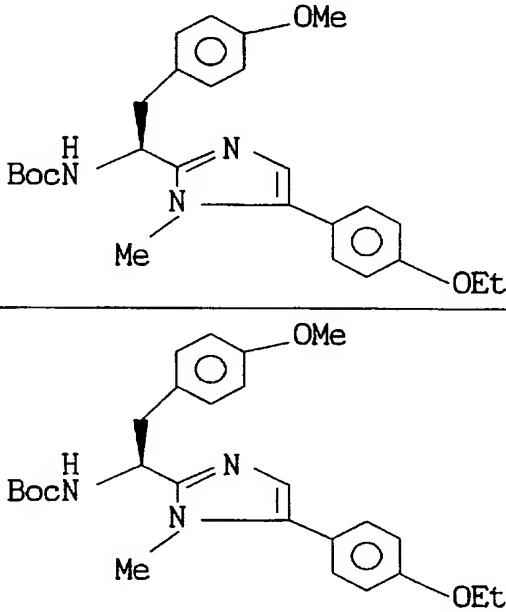
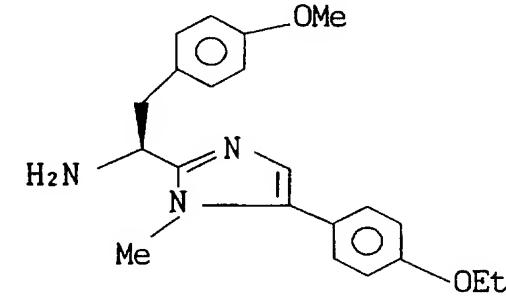
Table

Preparation No.	Formula
80	<p>Chemical structure of Preparation 80: 2-(4-fluorophenyl)propanamide hydrochloride. It features a 4-fluorophenyl ring connected to a propanoyl group (-CH₂COOH), which is also linked to an amino group (-NH₂) and a hydrogen chloride counterion (-HCl).</p>
81	<p>Chemical structure of Preparation 81: 2-(4-fluorophenyl)-N-(2-methoxyphenyl)propanamide. It features a 4-fluorophenyl ring connected to a propanoyl group (-CH₂COOH), which is further substituted with an N-(2-methoxyphenyl)amino group.</p>
81	<p>Chemical structure of Preparation 81: 2-(4-fluorophenyl)-N-(2-methoxyphenyl)-N-(2-methylimidazol-1-yl)propanamide. It features a 4-fluorophenyl ring connected to a propanoyl group (-CH₂COOH), which is further substituted with an N-(2-methoxyphenyl)amino group and an N-(2-methylimidazol-1-yl)amino group.</p>

Table

Preparation No.	Formula
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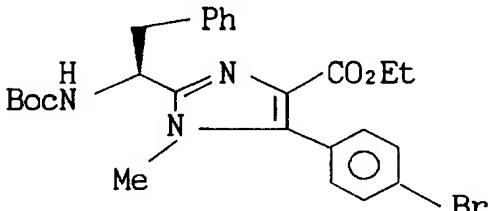
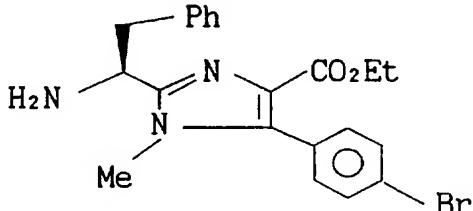
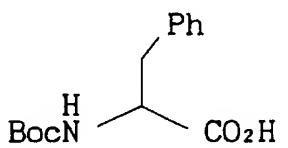
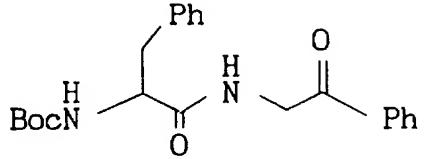
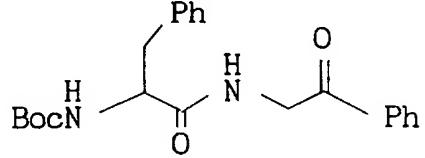
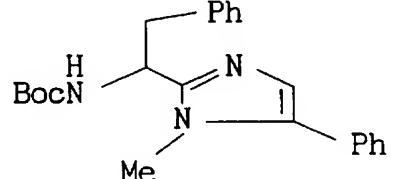
Table

Preparation No.	Formula
84	 <p>Chemical structure of Preparation 84: A chiral amide where a central carbon atom is bonded to a phenyl ring substituted with an OMe group, a CONH group, a methyl group, and a hydrogen atom. A vertical wedge bond is shown between the central carbon and the CONH group.</p>
85	 <p>Chemical structure of Preparation 85: A pyrimidine derivative where a central carbon atom is bonded to a phenyl ring substituted with an OMe group, a CONH group, a methyl group, and a hydrogen atom. The pyrimidine ring has a 4-methyl group and a 2-substituted phenyl ring attached at the 5-position. A vertical wedge bond is shown between the central carbon and the CONH group.</p>
	 <p>Chemical structure of Preparation 85: A pyrimidine derivative where a central carbon atom is bonded to a phenyl ring substituted with an OMe group, an H2N group, a methyl group, and a hydrogen atom. The pyrimidine ring has a 4-methyl group and a 2-substituted phenyl ring attached at the 5-position. A vertical wedge bond is shown between the central carbon and the H2N group.</p>

Table

Preparation No.	Formula
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86	
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Table

Preparation No.	Formula
89	
	
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91	
	

Table

Preparation No.	Formula
92	<p>Chemical structure of compound 92: A pyrazine ring substituted with a phenyl group at position 2, a 2-methylpropyl group at position 3, and a 2-phenylpropyl group at position 4.</p>
	<p>Chemical structure of compound 92 derivative: Similar to compound 92, but the 2-methylpropyl group is replaced by an amino group (H₂N).</p>
93	<p>Chemical structure of compound 93: A chiral center with a phenyl ring substituted with an ethoxy group (OEt) and a Boc-protected amine group (-NH-Boc).</p>
	<p>Chemical structure of compound 93 derivative: Similar to compound 93, but the Boc-protected amine group is replaced by a carbamoyl group (-NH-C(=O)-CH₂-Br).</p>

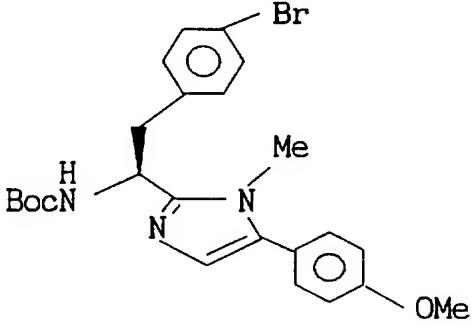
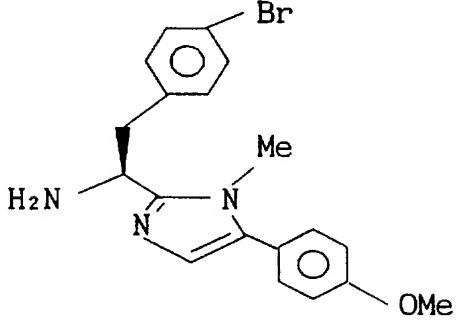
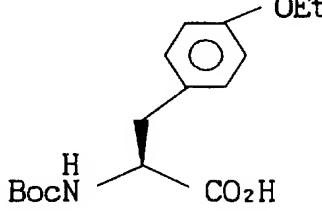
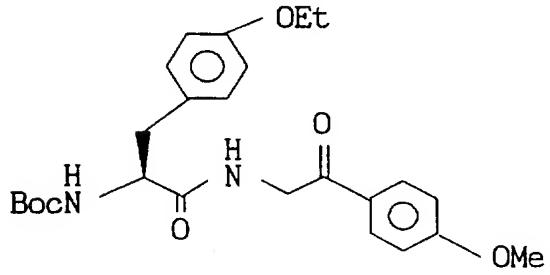
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Preparation No.	Formula
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Table

Preparation No.	Formula
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97	

Table

Preparation No.	Formula
98	 <p>Chemical structure of compound 98: A pyrazine ring substituted at position 2 with a 4-bromo-2-methoxyphenyl group, at position 4 with a methyl group, and at position 6 with a 4-methoxyphenyl group. A chiral center is shown with a wedge bond to the BocN group and a dash bond to the H atom.</p>
99	 <p>Chemical structure of compound 99: A pyrazine ring substituted at position 2 with a 4-bromo-2-methoxyphenyl group, at position 4 with a methyl group, and at position 6 with a 4-methoxyphenyl group. A chiral center is shown with a wedge bond to the H2N group and a dash bond to the Me group.</p>
99	 <p>Chemical structure of compound 99: A pyrazine ring substituted at position 2 with a 4-ethoxyphenyl group, at position 4 with a methyl group, and at position 6 with a 4-methoxyphenyl group. A chiral center is shown with a wedge bond to the BocN group and a dash bond to the CO₂H group.</p>
99	 <p>Chemical structure of compound 99: A pyrazine ring substituted at position 2 with a 4-ethoxyphenyl group, at position 4 with a methyl group, and at position 6 with a 4-methoxyphenyl group. A chiral center is shown with a wedge bond to the BocN group, a dash bond to the carbonyl group, and a hydrogen atom bonded to the nitrogen atom.</p>

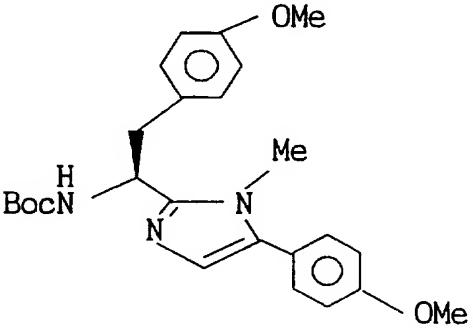
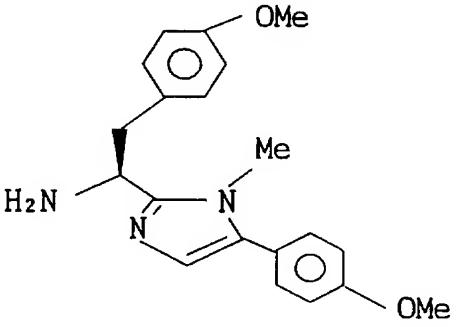
Table

Preparation No.	Formula
100	<p>Chemical structure of Preparation 100: A chiral amine derivative. It features a central carbon atom bonded to a phenyl ring substituted with an OEt group, a BocN group, a hydrogen atom, and an amino group (-NH-). The amino group is further bonded to a phenyl ring substituted with an OMe group.</p>
	<p>Chemical structure of Preparation 100: A chiral amine derivative. It features a central carbon atom bonded to a phenyl ring substituted with an OEt group, a BocN group, a hydrogen atom, and an amino group (-NH-). The amino group is further bonded to a phenyl ring substituted with an OMe group.</p>
101	<p>Chemical structure of Preparation 101: A chiral amine derivative. It features a central carbon atom bonded to a phenyl ring substituted with an OEt group, a BocN group, a hydrogen atom, and an amino group (-NH-). The amino group is further bonded to a phenyl ring substituted with an OMe group.</p>
	<p>Chemical structure of Preparation 101: A chiral amine derivative. It features a central carbon atom bonded to a phenyl ring substituted with an OEt group, a BocN group, a hydrogen atom, and an amino group (-NH-). The amino group is further bonded to a phenyl ring substituted with an OMe group.</p>

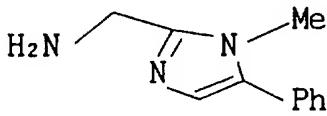
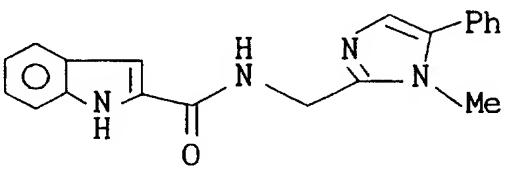
Table

Preparation No.	Formula
102	
103	

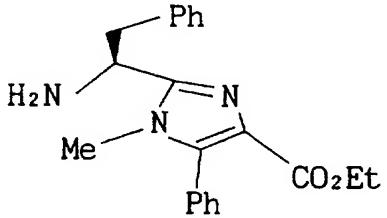
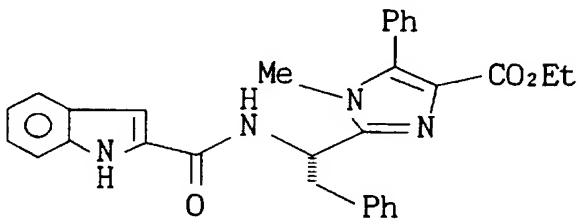
Table

Preparation No.	Formula
104	
	

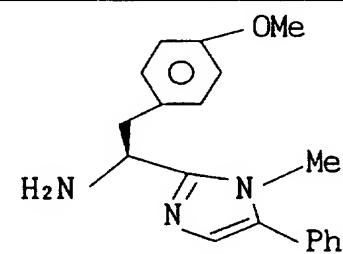
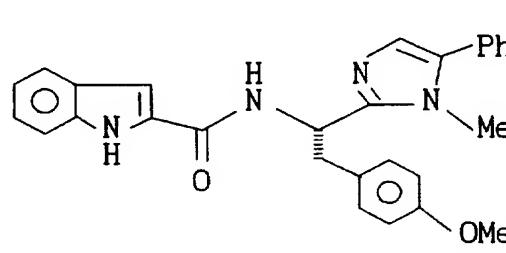
Table

Example No.	Formula
1	 <chem>CN(CC)C1=CN=C(C=C1)C2=CC=CC=C2</chem>
2	 <chem>CN(CC)C1=CN=C(C=C1)C2=CC=CC=C2=O</chem>

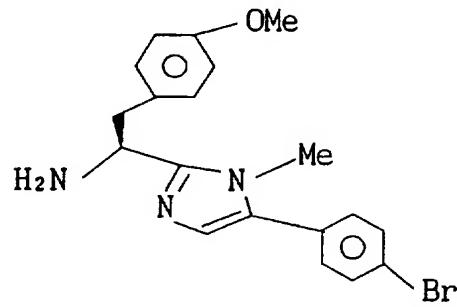
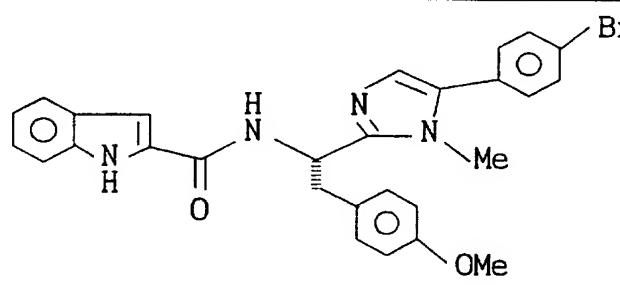
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Example No.	Formula
3	
4	

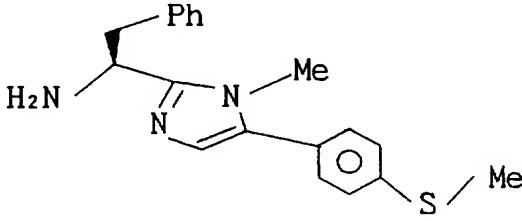
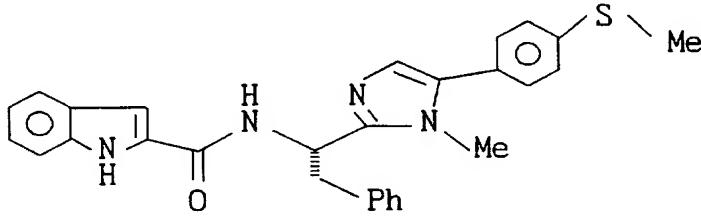
Table

Example No.	Formula
5	 <p>Chemical structure of compound 5: A pyrazine ring substituted at position 2 with a 4-methoxyphenyl group and at position 4 with a 2-amino-2-(4-methoxyphenyl)ethyl group.</p>
6	 <p>Chemical structure of compound 6: A pyrazine ring substituted at position 2 with a 4-methoxyphenyl group and at position 4 with a 2-amino-2-(4-methoxyphenyl)ethyl group, connected via a methylene bridge to a 2-hydroxy-2H-chromene-3-carbonyl group.</p>

Table

Example No.	Formula
7	 <p>Chemical structure of compound 7: A pyrazine ring substituted at position 2 with a 4-methoxyphenyl group and at position 4 with a 4-bromophenyl group. A 2-aminoethyl side chain is attached to the ring via a chiral center.</p>
8	 <p>Chemical structure of compound 8: A pyrazine ring substituted at position 2 with a 4-methoxyphenyl group and at position 4 with a 4-bromophenyl group. A 2-aminoethyl side chain is attached to the ring via a chiral center. This structure is identical to compound 7.</p>

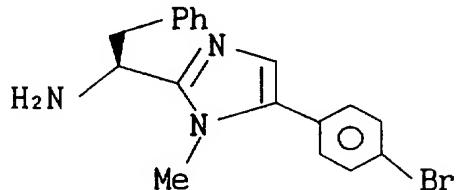
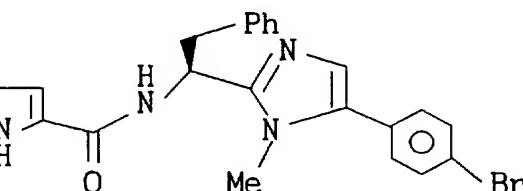
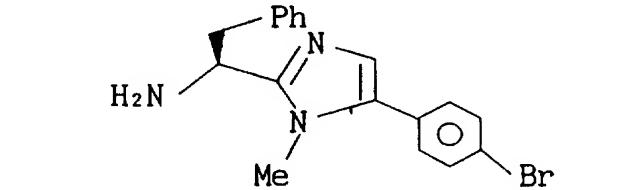
Table

Example No.	Formula
9	 <chem>CN1C=CC=C1c2ccccc2S(=O)(=O)C</chem>
10	 <chem>CC(=O)c1ccccc1N2Cc3ccccc3S(=O)(=O)C2</chem>

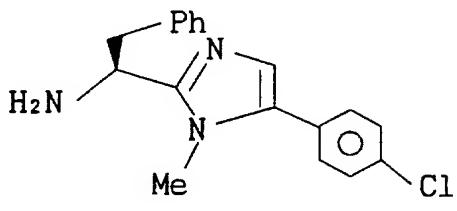
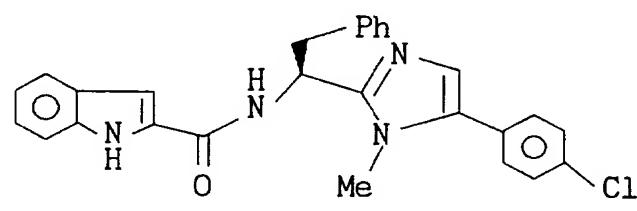
Table

Example No.	Formula
11	<p>Chemical structure of compound 11: A pyrazine ring substituted with an amino group (H₂N) at position 2, a chlorine atom (Cl) at position 4, and a 4-bromo-2-chlorophenyl group at position 6.</p>
12	<p>Chemical structure of compound 12: A tricyclic compound consisting of a benzodioxole ring fused to a pyrazine ring, which is further substituted with an amino group (H₂N) at position 2, a chlorine atom (Cl) at position 4, and a 4-bromo-2-chlorophenyl group at position 6.</p>

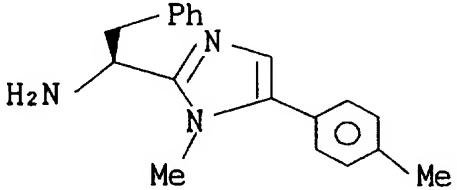
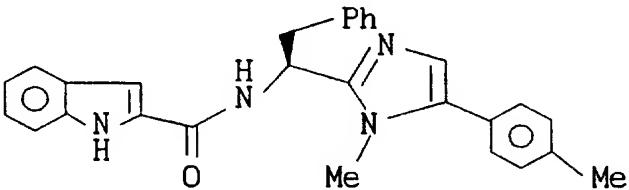
Table

Example No.	Formula
13	
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14	

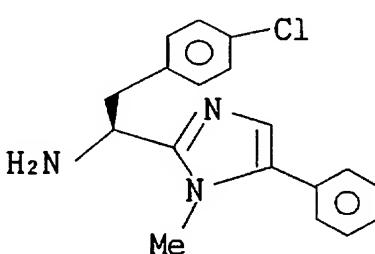
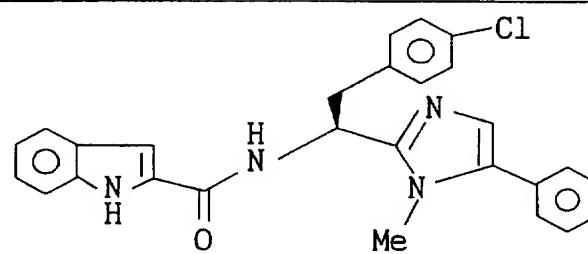
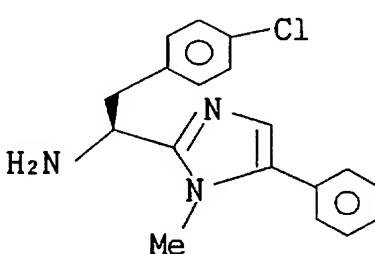
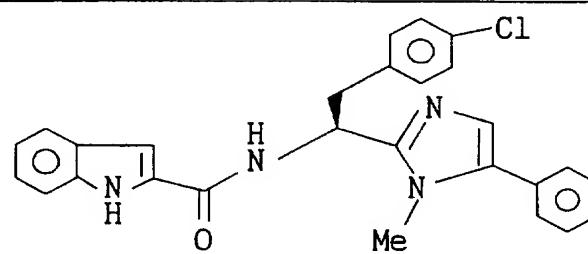
Table

Example No.	Formula
15	
16	

Table

Example No.	Formula
17	
18	

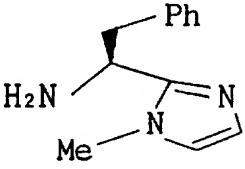
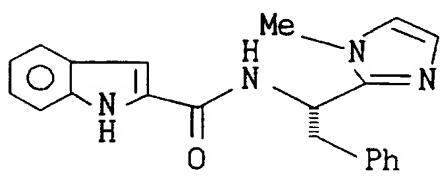
Table

Example No.	Formula
19	
20	
19	
20	

Table

Example No.	Formula
21	<p>Chemical structure of compound 21: A pyrazine ring substituted at position 2 with a 4-(4-methylphenyl)butyl group and at position 4 with an (S)-1-aminocyclopropylmethyl group.</p>
22	<p>Chemical structure of compound 22: A pyrazine ring substituted at position 2 with a 4-(4-methylphenyl)butyl group and at position 4 with an (R)-1-aminocyclopropylmethyl group.</p>

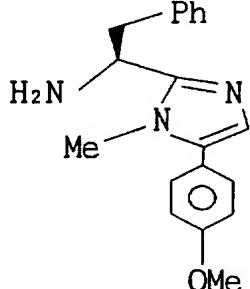
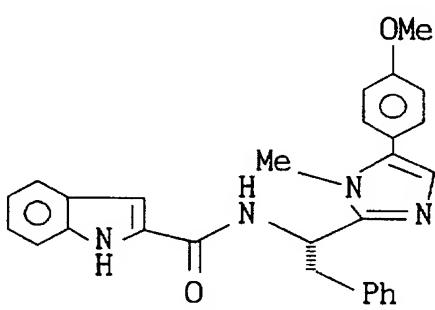
Table

Example No.	Formula
23	
24	

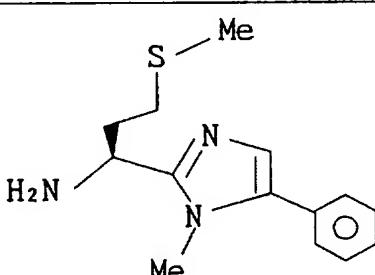
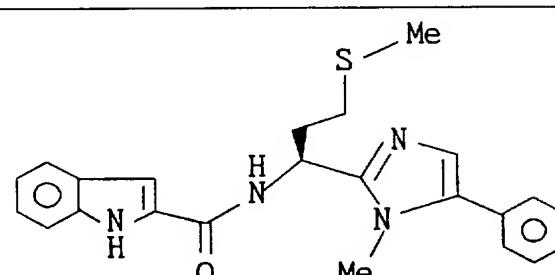
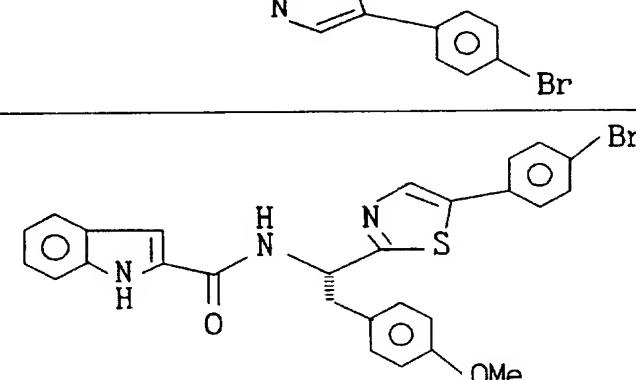
Table

Example No.	Formula
25	<p>Chemical structure of compound 25: 1-(2-((2-methyl-1-phenylpropyl)amino)-2-phenylpropyl)-4-quinolone. It features a quinolone core substituted at position 4 with a 2-aminobutyl group. The butyl chain has a phenyl ring attached to the second carbon and a methyl group at the first carbon. The third carbon of the butyl chain is substituted with a 2-phenylpropyl group.</p>
26	<p>Chemical structure of compound 26: 1-(2-((2-methyl-1-phenylpropyl)amino)-2-phenylpropyl)-4-(2-phenyl-1-phenylpropyl)quinolone. It consists of a quinolone core substituted at position 4 with a 2-aminobutyl group. The butyl chain has a phenyl ring attached to the second carbon and a methyl group at the first carbon. The third carbon of the butyl chain is substituted with another 2-aminobutyl group, which has a phenyl ring attached to the second carbon and a methyl group at the first carbon.</p>

Table

Example No.	Formula
27	
28	

Table

Example No.	Formula
29	 The structure shows a 5-methylimidazole ring system. At position 2, there is a phenyl group. At position 4, there is a methyl group. At position 5, there is a cyclopropylmethyl group substituted with a methylthio group (-S-Me) and an amino group (-NH ₂). A chiral center is indicated with a wedge bond.
30	 The structure shows a cyclopropane ring substituted with an amine group (-NH ₂) at one end and a 2-methylthio group at the other. This ring is further substituted with a 4-bromophenyl group and a 2-methoxyphenyl group. A chiral center is indicated with a wedge bond.
	 The structure shows a cyclopropane ring substituted with an amine group (-NH ₂) at one end and a 2-methylthio group at the other. This ring is further substituted with a 4-bromophenyl group and a 2-methoxyphenyl group. A chiral center is indicated with a wedge bond.

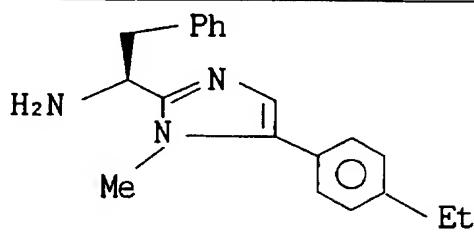
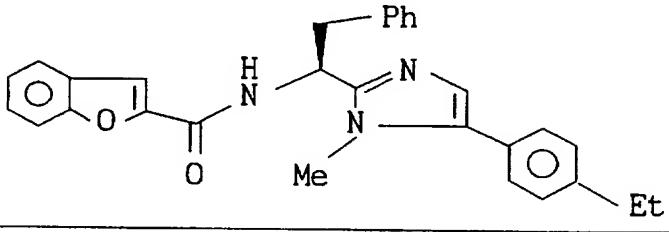
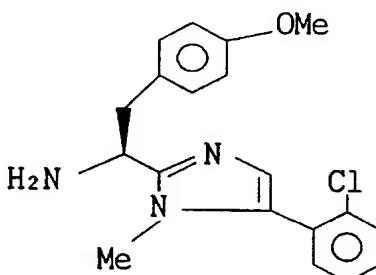
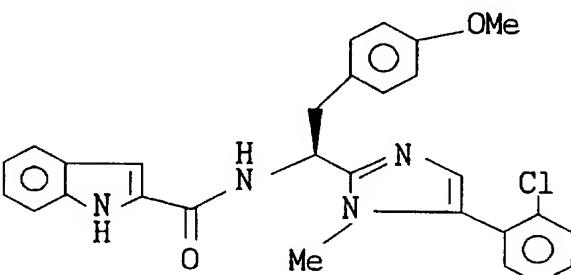
Table

Example No.	Formula
31	<p>Chemical structure of compound 31: A chiral center with an amino group (H_2N) and a phenyl ring substituted with a methoxy group (OMe). It is connected to a thiophene ring which has a double bond to a phenyl ring substituted with a bromine atom (Br).</p>
	<p>Chemical structure of compound 31 derivative: A complex molecule featuring a thiomorpholine-4-carbonyl group, a thiophene ring, and a phenyl ring substituted with a bromine atom (Br).</p>
32	<p>Chemical structure of compound 32: A chiral center with an amino group (H_2N) and a phenyl ring substituted with a methyl group (Me). It is connected to a pyrimidine ring which has a double bond to a phenyl ring substituted with an ethoxy group (OEt).</p>
	<p>Chemical structure of compound 32 derivative: A complex molecule featuring a thiomorpholine-4-carbonyl group, a pyrimidine ring, and a phenyl ring substituted with an ethoxy group (OEt).</p>

Table

Example No.	Formula
33	<p>Chemical structure of compound 33: A pyrazine ring substituted with an amino group (H₂N) at position 2, a methyl group (Me) at position 4, and a 2-(4-ethoxyphenyl)-1-phenylethyl group at position 6.</p>
	<p>Chemical structure of compound 33 derivative: Similar to compound 33, but the 4-ethoxyphenyl group is replaced by a 2-(4-acetylphenyl)-1-phenylethyl group.</p>
34	<p>Chemical structure of compound 34: A pyrazine ring substituted with an amino group (H₂N) at position 2, a methyl group (Me) at position 4, and a 2-(4-ethylphenyl)-1-phenylethyl group at position 6.</p>
	<p>Chemical structure of compound 34 derivative: Similar to compound 34, but the 4-ethylphenyl group is replaced by a 2-(4-acetylphenyl)-1-phenylethyl group.</p>

Table

Example No.	Formula
35	 
36	 

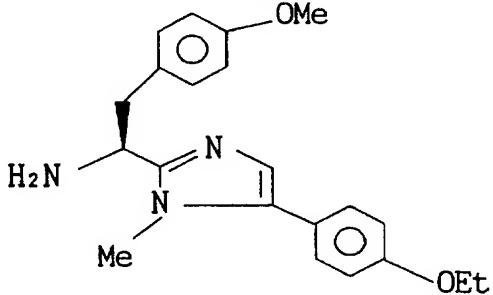
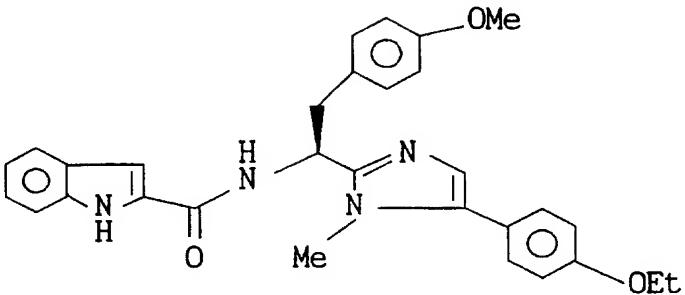
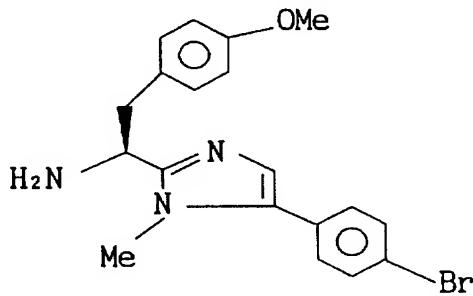
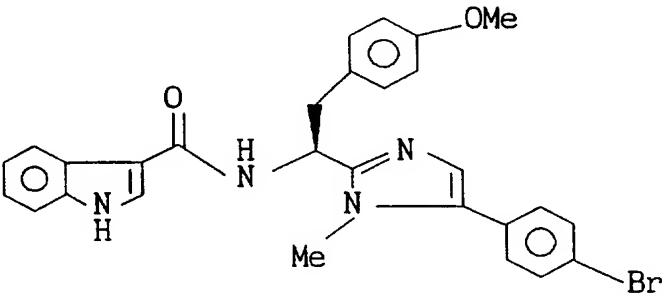
Table

Example No.	Formula
37	
38	

Table

Example No.	Formula
39	
40	

Table

Example No.	Formula
41	
	
42	
	

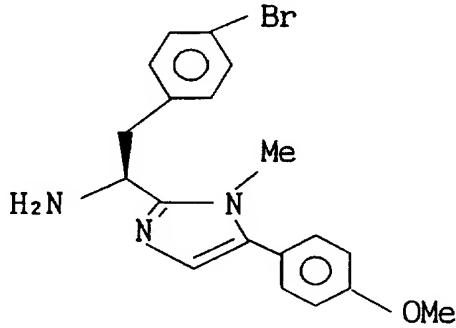
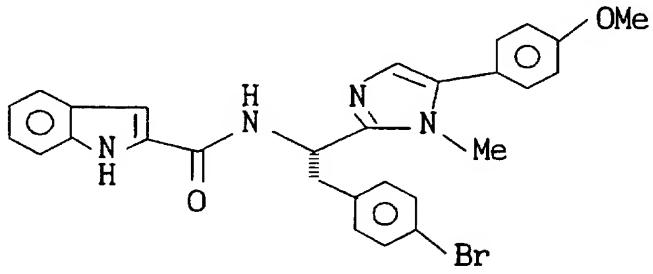
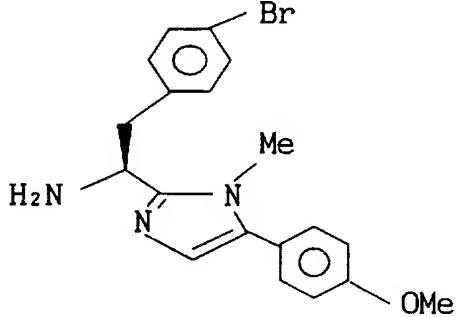
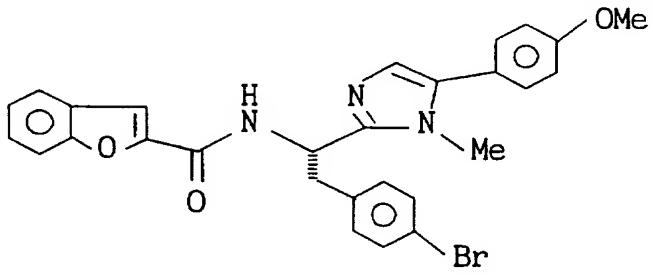
Table

Example No.	Formula
43	<p>Chemical structure of compound 43: A pyrazine ring with an amino group at position 2 and a methyl group at position 4. At position 6, there is a 2-(ethoxycarbonyl)-5-(4-bromophenyl)ethyl group attached via a chiral center.</p>
	<p>Chemical structure of compound 43 derivative: Similar to compound 43, but the 2-(ethoxycarbonyl)-5-(4-bromophenyl)ethyl group is attached to the pyrazine ring via a different linkage, likely a carbonyl group.</p>
44	<p>Chemical structure of compound 44: A pyrazine ring with an amino group at position 2 and a methyl group at position 4. At position 6, there is a phenyl group attached.</p>
	<p>Chemical structure of compound 44 derivative: Similar to compound 44, but the phenyl group is attached to the pyrazine ring via a chiral center.</p>

Table

Example No.	Formula
45	
46	

Table

Example No.	Formula
47	
	
48	
	

Table

Example No.	Formula
49	
50	
50	

Preparation 1

To an ice-cooled mixture of N-(tert-butoxycarbonyl)glycine (1.40 g) and 2-aminoacetophenone hydrochloride (1.61 g) in dichloromethane (14 ml) was added 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide (1.49 g). The mixture was stirred at room temperature for 12 hours. A saturated aqueous sodium hydrogencarbonate solution was added to the mixture, and then the mixture was extracted three times with chloroform. The organic layer was washed with brine, dried over magnesium sulfate, filtered and concentrated. The residue was purified by column chromatography (silica gel, chloroform/methanol=40/1) to give the object compound as white powder (689 mg).

MASS (ESI) (m/z) : 293 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.47(9H,s), 3.92(2H,d,J=5Hz), 4.78(2H,s), 5.13(1H,br s), 7.05(1H,br s), 7.45-7.70(3H,m), 7.92-8.04(2H,m)

Preparation 2

A solution of the starting compound (669 mg) and 40% methylamine (0.7 ml) in a mixture of acetic acid (0.7 ml) and xylene (7 ml) was refluxed for 4 hours in a flask equipped with a Dean-Stark trap. The mixture was concentrated, neutralized with 1N hydroxide solution, and extracted three times with chloroform. The organic layer was washed with brine, dried over magnesium sulfate, filtered and concentrated. The residue was purified by column chromatography (silica gel, chloroform/methanol=50/1) to give the object compound as an oil (445 mg).

MASS (ESI) (m/z) : 288 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.46(9H,s), 3.60(3H,s), 4.48(2H,d,J=5Hz), 5.33(1H,br s), 6.99(1H,s), 7.30-7.52(5H,m)

Preparation 3

The starting compound (430 mg) was dissolved in trifluoroacetic acid (1.5 ml) and the mixture was stirred at room temperature for 1 hour. The mixture was concentrated, made basic with 1N sodium

hydroxide solution and extracted three times with chloroform. The organic layer was dried over magnesium sulfate and filtered. Evaporation of the solvent gave the object compound as an oil (314 mg).

MASS (ESI) (m/z) : 188 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.57(3H,s), 3.98(2H,s), 6.98(1H,s), 7.26-7.50(5H,m)

Preparation 4

To a solution of the starting compound (2.12 g) in tetrahydrofuran (20 ml) was added successively isobutyl chloroformate (1.1 ml) and N-methylmorpholine (0.9 ml) at -25°C, and the mixture was stirred at the temperature for 5 minutes. The above mixture was added to a solution of dl-2-benzoylglycine ethyl ester hydrochloride (2.05 g) and N-methylmorpholine (0.9 ml) in tetrahydrofuran (5 ml) at -20°C, and the mixture was allowed to warm to room temperature for 2 hours. Water was added to the mixture, and the mixture was extracted three times with chloroform. The organic layer was washed with brine, dried over magnesium sulfate, filtered, and concentrated. The residue was purified by column chromatography (silica gel, hexane/ethyl acetate=3/1) to give the object compound as an oil (2.36 g).

MASS (ESI) (m/z) : 455 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.13(3H,t,J=7Hz), 1.41(9H,s), 2.95-3.21(2H,m), 4.13(2H,q,J=7Hz), 4.38-4.60(1H,m), 4.83-5.05(1H,m), 6.02-6.20(1H,m), 7.10-7.37(6H,m), 7.42-7.71(3H,m), 8.01-8.18(2H,m)

Preparation 5

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 450 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.12(3H,t,J=7Hz), 1.40(9H,s), 2.68(3H,s), 3.08-3.42(2H,m), 4.21(2H,q,J=7Hz), 4.89-5.05(1H,m), 5.77(1H,br d,J=8Hz), 6.96-7.48(10H,m)

Preparation 6

The object compound was obtained according to a similar manner to that of Preparation 3.

MASS (ESI) (m/z) : 350 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.08(3H, t, J=7Hz), 2.80(3H, s), 3.21-3.48(2H, m), 4.15(2H, q, J=7Hz), 4.25-4.72(3H, m), 7.00-7.48(10H, m)

Preparation 7

The object compound was obtained according to a similar manner to that of Preparation 1.

MASS (ESI) (m/z) : 413 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.41(9H, s), 3.05(2H, d, J=6Hz), 3.75(3H, s), 4.43(1H, br s), 4.58-4.81(2H, m), 5.05(1H, br s), 6.81(2H, d, J=8Hz), 6.91(1H, br s), 7.12(2H, d, J=8Hz), 7.42-7.68(3H, m), 7.95(2H, d, J=7Hz)

Preparation 8

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 408 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.42(9H, s), 3.00-3.33(2H, m), 3.02(3H, s), 3.77(3H, s), 4.89-5.04(1H, m), 5.63(1H, d, J=8Hz), 6.76(2H, d, J=8Hz), 6.94(2H, d, J=8Hz), 7.02(1H, s), 7.18-7.45(5H, m)

Preparation 9

To a solution of the starting compound (3.10 g) in methanol (15 ml) was added concentrated hydrochloric acid (3 ml), and the mixture was heated to 50°C for 2 hours. The mixture was concentrated, made basic with a 1N sodium hydroxide solution, and extracted three times with chloroform. The organic layer was dried over magnesium sulfate, and filtered. Evaporation of the solvent gave the object compound (2.35 g).

MASS (ESI) (m/z) : 308 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.02-3.22(2H, m), 3.21(3H, s),
3.78(3H, s), 4.11(1H, t, J=7Hz), 6.81(2H, d, J=8Hz),
6.99(2H, d, J=8Hz), 7.04(1H, s), 7.21-7.48(5H, m)

Preparation 10

The object compound was obtained according to a similar manner to that of Preparation 1.

MASS (ESI) (m/z) : 491, 493 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.41(9H, s), 3.04(2H, d, J=6Hz),
3.75(3H, s), 4.42(1H, br s), 4.54-4.77(2H, m), 5.00(1H, br s),
6.81(2H, d, J=8Hz), 6.85(1H, br s), 7.12(2H, d, J=8Hz),
7.63(2H, d, J=7Hz), 7.80(2H, d, J=7Hz)

Preparation 11

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 486, 488 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.41(9H, s), 3.00(3H, s),
3.01-3.32(2H, m), 3.76(3H, s), 4.88-5.02(1H, m),
5.57(1H, d, J=8Hz), 6.76(2H, d, J=8Hz), 6.88-7.18(5H, m),
7.51(2H, d, J=8Hz)

Preparation 12

The object compound was obtained according to a similar manner to that of Preparation 9.

MASS (ESI) (m/z) : 386, 388 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.02-3.18(2H, m), 3.20(3H, s),
3.78(3H, s), 4.12(1H, t, J=7Hz), 6.81(2H, d, J=8Hz),
6.98(2H, d, J=8Hz), 7.03(1H, s), 7.15(2H, d, J=8Hz),
7.52(2H, d, J=8Hz)

Preparation 13

The object compound was obtained according to a similar manner to that of Preparation 1.

MASS (ESI) (m/z) : 429 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.41(9H, s), 2.52(3H, s),

2.99-3.21(2H,m), 4.48(1H,br s), 4.53-4.79(2H,m),
5.03(1H,br s), 6.90(1H,br s), 7.13-7.25(7H,m),
7.83(2H,d,J=8Hz)

Preparation 14

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 424 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 1.40(9H,s), 2.50(3H,s),
2.94(3H,s), 3.00-3.40(2H,m), 4.90-5.10(1H,m),
5.59(1H,br d,J=8Hz), 6.95-7.35(10H,m)

Preparation 15

The object compound was obtained according to a similar manner to that of Preparation 9.

MASS (ESI) (m/z) : 324 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 2.50(3H,s), 3.08-3.27(2H,m),
3.17(3H,s), 4.16(1H,t,J=7Hz), 7.03(1H,s), 7.05-7.35(9H,m)

Preparation 16

The object compound was obtained according to a similar manner to that of Preparation 1.

MASS (ESI) (m/z) : 495, 497 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 1.40(9H,s), 2.98-3.20(2H,m),
4.47(1H,m), 4.55-4.78(2H,m), 5.10(1H,br d,J=8Hz),
7.01(1H,br s), 7.14(2H,d,J=8Hz), 7.25(2H,d,J=8Hz),
7.64(2H,d,J=8Hz), 7.81(2H,d,J=8Hz)

Preparation 17

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 490, 492 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 1.39(9H,s), 3.12(3H,s),
3.13-3.22(2H,m), 4.91-5.08(1H,m), 5.47(1H,br d,J=9Hz),
6.90-7.30(7H,m), 7.52(2H,d,J=8Hz)

Preparation 18

The object compound was obtained according to a similar manner to that of Preparation 9.

MASS (ESI) (m/z) : 390, 392 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.02-3.26(2H,m), 3.27(3H,s), 4.11(1H,t,J=7Hz), 7.02(2H,d,J=8Hz), 7.03(1H,s), 7.15(2H,d,J=8Hz), 7.22(2H,d,J=8Hz), 7.53(2H,d,J=8Hz)

Preparation 19

The object compound was obtained according to a similar manner to that of Preparation 1.

amorphous solid

MASS : 461 (M+1)

¹H-NMR (CDCl₃) δ : 1.39(9H,s), 3.00-3.20(2H,m), 4.40-4.78(3H,m), 5.03(1H,bs), 6.89(1H,bs), 7.19-7.38(5H,m), 7.63(2H,d,J=8Hz), 7.82(2H,d,J=8Hz)

Preparation 20

The object compound was obtained according to a similar manner to that of Preparation 2.

mp : 162-164°C

MASS : 456 (M+1)

¹H-NMR (CDCl₃) δ : 1.41(9H,s), 2.97(3H,s), 3.11(1 x 1/3H,d,J=8Hz), 3.15(1 x 2/3H,d,J=8Hz), 3.31(1 x 2/3H,d,J=8Hz), 3.35(1 x 1/3H,d,J=8Hz), 4.91-5.08(1H,m), 5.59(1H,d,J=8Hz), 6.99-7.07(3H,m), 7.09(2H,d,J=8Hz), 7.18-7.23(3H,m), 7.51(2H,d,J=8Hz)

Preparation 21

The object compound was obtained according to a similar manner to that of Preparation 3.

oil

MASS : 356 (M+1)

¹H-NMR (CDCl₃) δ : 3.10-3.25(2H,m), 3.20(3H,s), 4.17(1H,t,J=8Hz), 7.05(1H,s), 7.10(2H,d,J=8Hz), 7.14(2H,d,J=8Hz), 7.20-7.32(3H,m), 7.53(2H,d,J=8Hz)

Preparation 22

The object compound was obtained according to a similar manner to that of Preparation 1.

amorphous solid

MASS : 417 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ : 1.40(9H,s), 3.11(2H,d,J=8Hz),
4.40-4.60(1H,m), 4.60-4.78(2H,m), 5.00(1H,bs), 6.84(1H,bs),
7.17-7.36(5H,m), 7.49(2H,d,J=8Hz), 7.90(2H,d,J=8Hz)

Preparation 23

The object compound was obtained according to a similar manner to that of Preparation 2.

amorphous solid

MASS : 412 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ : 1.41(9H,s), 2.92(3H,s), 3.00-3.20(1H,m),
3.24-3.40(1H,m), 5.00(1H,q,J=8Hz), 5.59(1H,d,J=8Hz),
7.00-7.10(3H,m), 7.14(2H,d,J=8Hz), 7.18-7.30(3H,m),
7.37(2H,d,J=8Hz)

Preparation 24

The object compound was obtained according to a similar manner to that of Preparation 3.

oil

MASS : 312 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ : 3.10-3.28(2H,m), 3.18(3H,s),
4.10-4.24(1H,m), 7.08(2H,d,J=8Hz), 7.11(1H,s),
7.21(2H,d,J=8Hz), 7.22-7.33(3H,m), 7.39(2H,d,J=8Hz)

Preparation 25

The object compound was obtained according to a similar manner to that of Preparation 1.

mp : 135-139°C

MASS : 397 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ : 1.41(9H,s), 2.41(3H,s), 3.00-3.20(2H,m),
4.50(1H,d,J=5Hz), 4.57-4.78(2H,m), 5.07(1H,d,J=5Hz),

6.91(1H,s), 7.18-7.33(7H,m), 7.83(2H,d,J=8Hz)

Preparation 26

The object compound was obtained according to a similar manner to that of Preparation 2.

mp : 131-133°C

MASS : 392 (M+1)

¹H-NMR (CDCl₃) δ : 1.39(9H,s), 2.38(3H,s), 2.97(3H,s),
3.11(1 x 1/3H,d,J=8Hz), 3.17(1 x 2/3H,d,J=8Hz),
3.31(1 x 2/3H,d,J=8Hz), 3.36(1 x 1/3H,d,J=8Hz),
4.93-5.08(1H,m), 5.59(1H,d,J=8Hz), 7.00(1H,s),
7.01-7.09(2H,m), 7.09-7.16(2H,m), 7.16-7.28(5H,m)

Preparation 27

The object compound was obtained according to a similar manner to that of Preparation 3.

oil

MASS : 292 (M+1)

¹H-NMR (CDCl₃) δ : 2.37(3H,s), 3.10-3.27(2H,m), 3.19(3H,s),
4.17(1H,t,J=8Hz), 7.01(1H,s), 7.09(2H,d,J=8Hz),
7.12-7.33(7H,m)

Preparation 28

To an ice-cooled mixture of the starting compound (599 mg), 2-aminoacetophenone hydrochloride (362 mg) and 1-hydroxybenzotriazole (270 mg) in dichloromethane (6 ml) was added 1-(3-dimethylamino-propyl)-3-ethylcarbodiimide (349 mg). The mixture was stirred at room temperature for 12 hours. A saturated aqueous sodium hydrogencarbonate solution was added to the mixture, and then the mixture was extracted three times with chloroform. The organic layer was washed with brine, dried over magnesium sulfate, filtered, and concentrated. The residue was purified by column chromatography (silica gel, chloroform/methanol=70/1) to give the object compound (823 mg).

MASS (ESI) (m/z) : 417 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.41(9H,s), 2.96-3.20(2H,m),
4.47(1H,m), 4.70(2H,AB of ABX,J_{A,B}=15Hz), 5.01(1H,br s),
6.92(1H,br s), 7.13(2H,d,J=8Hz), 7.24(2H,d,J=8Hz),
7.41-7.68(3H,m), 7.88-8.00(2H,m)

Preparation 29

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 412 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.40(9H,s), 3.13(3H,s),
3.15-3.32(2H,m), 4.92-5.07(1H,m), 5.58(1H,br d,J=8Hz),
6.93-7.55(10H,m)

Preparation 30

The object compound was obtained according to a similar manner to that of Preparation 3.

MASS (ESI) (m/z) : 312 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.06-3.25(2H,m), 3.24(3H,s),
4.17(1H,t,J=7Hz), 6.98-7.50(10H,m)

Preparation 31

The starting compound (1.1 g) and glyoxal trimeric dihydrate (930 mg) were stirred in methanol (7 ml) at -10°C. Ammonia was bubbled through the solution for 5 minutes and the mixture was stirred at -10°C for 1 hour. The mixture was allowed to warm to room temperature over 18 hours, then poured into water, and extracted twice with dichloromethane. The combined extracts was dried over anhydrous magnesium sulfate and concentrated in vacuo. The residue was purified by flash column chromatography over silica gel with a dichloromethane-methanol gradient (20:1 and 10:1) as eluent to give the object compound as an off-white solid (698.6 mg).

mp : 180.5-184°C

MASS : 288 (M+H)⁺

¹H-NMR (CDCl₃) δ : 1.40(9H,s), 3.29(2H,d,J=7.5Hz),
4.90(1H,q,J=7.5Hz), 5.25(1H,bs,J=7.5Hz), 6.89(1H,bs),

6.99(1H,bs), 7.12(2H,d,J=7.5Hz), 7.18-7.30(3H,m),
9.78(1H,bs)

Preparation 32

To a precooled solution of the starting compound (500 mg) in N,N-dimethylformamide (5 ml) was added 85% potassium hydroxide powder (115 mg). After the mixture was stirred for 1 hour on an ice bath, α -chloro-p-xylene (230.4 μ l) was added dropwise to the reaction mixture. The resulting suspension was stirred at 5°C for 14 hours, then poured into water, and extracted with chloroform. The organic layer was washed twice with water and a saturated sodium chloride solution, dried over anhydrous magnesium sulfate, and concentrated in vacuo. The residue was washed with diethyl ether to give the object compound as a colorless solid (418.3 mg).

mp : 157-158.5°C

MASS : 392 (M+H)⁺

¹H-NMR (CDCl₃) δ : 1.36(9H,s), 2.30(3H,s), 3.19(2H,m),
4.63(1H,d,J=16.0Hz), 4.71(1H,d,J=16.0Hz), 5.01(1H,m),
5.32(1H,m), 6.63(1H,s), 6.77(2H,d,J=7.5Hz), 6.98-7.23(8H,m)

Preparation 33

The object compound was obtained according to a similar manner to that of Preparation 3.

colorless oil

MASS : 292 (M+H)⁺

¹H-NMR (CDCl₃) δ : 2.31(3H,s), 3.02(1H,dd,J=13.5 and 7.5Hz),
3.12(1H,dd,J=13.5 and 7.5Hz), 4.06(1H,t,J=7.5Hz),
4.76(1H,d,J=14.5Hz), 4.83(1H,d,J=14.5Hz), 6.71(1H,s),
6.86(2H,d,J=7.5Hz), 6.99-7.04(3H,m), 7.10(2H,d,J=7.5Hz),
7.20-7.30(3H,m)

Preparation 34

The object compound was obtained according to a similar manner to that of Preparation 1.

white crystals

mp : 134-135°C

MASS (ESI) (m/z) : 383 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 1.41(9H,s), 3.00-3.22(2H,m),
4.47(1H,m), 4.69(2H,AB of ABX, J_{AB}=19Hz),
5.03(1H,br s), 6.90(1H,br s), 7.16-7.68(8H,m),
7.95(2H,d,J=8Hz)

Preparation 35

The object compound was obtained according to a similar manner to that of Preparation 2.

white crystals

mp : 130-131°C

MASS (ESI) (m/z) : 378 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 1.41(9H,s), 2.96(3H,s),
3.06-3.20(1H,m), 3.28-3.40(1H,m), 4.92-5.06(1H,m),
5.57(1H,br d,J=9Hz), 7.00-7.43(11H,m)

Preparation 36

The object compound was obtained according to a similar manner to that of Preparation 3.

white powder

MASS (ESI) (m/z) : 278 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 3.10-3.28(2H,m), 3.18(3H,s),
4.16(1H,t,J=7Hz), 7.05(1H,s), 7.07-7.45(10H,m)

Preparation 37

The starting compound (600 mg) was heated at 40°C for 2 hours in methyl iodide (10 ml). The reaction mixture was evaporated, and the residue was suspended in an aqueous sodium carbonate solution. The mixture was extracted with chloroform. The organic layer was washed successively with water and a saturated sodium chloride solution, dried over anhydrous magnesium sulfate and concentrated in vacuo. The residue was purified by flash column chromatography over silica gel with a chloroform-methanol (20:1) as eluent to give the object compound as a pale yellow oily solid (376.5 mg).

mp : 116-119°C

MASS (ESI) (m/z) : 302 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.40(9H,s), 3.05(3H,s),
3.10(1H,dd,J=14.5, 9.0Hz), 3.29(1H,dd,J=14.5, 4.5Hz),
4.93(1H,m), 5.50(1H,br d,J=7.5Hz), 6.63(1H,s),
6.95-7.02(3H,m), 7.15-7.24(3H,m)

Preparation 38

The object compound was obtained according to a similar manner to that of Preparation 3.

yellow oil

MASS (ESI) (m/z) : 202 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.09(1H,dd,J=14.5, 7.5Hz),
3.13(1H,dd,J=14.5, 7.5Hz), 3.23(3H,s), 4.12(1H,t,J=7.5Hz),
6.69(1H,s), 6.99(1H,s), 7.03(2H,d,J=7.5Hz), 7.16-7.32(3H,m)

Preparation 39

The object compound was obtained according to a similar manner to that of Preparation 3.

yellow oil

MASS (ESI) (m/z) : 188 (M+H)⁺

¹H-NMR (CDCl₃, δ) 2.82(1H,dd,J=14.5, 8.5Hz),
3.37(1H,dd,J=14.5, 2.5Hz), 4.35(1H,dd,J=8.5, 2.5Hz),
6.99(2H,s), 7.12(2H,d,J=7.5Hz), 7.20-7.34(4H,m)

Preparation 40

A mixture of 6-acetylquinoline (2.0 g), hydroxylamine hydrochloride (1.0 g) and sodium carbonate (1.7 g) in ethanol (20 ml) was refluxed for 1 hour. After cooling to room temperature, water was added to the mixture. The precipitate was collected and washed with diethyl ether to give the object compound as a pale yellow solid (1.7 g).

mp : 170-173°C

MASS (ESI) (m/z) : 187 (M+H)⁺

¹H-NMR (CDCl₃, δ) 2.43(3H,s), 7.44(1H,dd,J=7.5, 4.5Hz),

8.00(1H,s), 8.16-8.23(3H,m), 8.94(1H,d,J=4.5Hz), 9.46(1H,s)

Preparation 41

To a solution of the starting compound (1.50 g) in pyridine (15 ml) cooled to 0°C was added p-toluenesulfonyl chloride (1.84 g) with stirring under an atmosphere of nitrogen, and the mixture was stirred at 0°C for 9 hours. After the reaction mixture was poured into ice-water, the precipitate was collected and washed successively with water and 2-propanol to give the object compound as a pale brown solid (1.62 g).

mp : 119.5-121°C

MASS (ESI) (m/z) : 341 (M+H)⁺

¹H-NMR (CDCl₃, δ) 2.43(3H,s), 2.48(3H,s),
7.36(2H,d,J=7.5Hz), 7.44(1H,dd,J=7.5, 4.5Hz), 7.92-8.03(4H,m),
8.07(1H,d,J=7.5Hz), 8.18(1H,d,J=7.5Hz), 8.95(1H,d,J=4.5Hz)

Preparation 42

Potassium (258.4 mg) was added to a suspension of the starting compound (1.5 g) in ethanol (40 ml), and the mixture was stirred at room temperature for 72 hours. The precipitate of potassium p-toluenesulfonate was removed by filtration, and the filtrate was diluted with diethyl ether (400 ml). A further precipitate of the potassium salt was filtered off, and the ethereal solution was extracted twice with 1.5N hydrochloric acid (50 ml). The combined extracts were evaporated in vacuo, and the residue was recrystallized from 2-propanol to give the object compound as an off-white solid (1.31 g).

mp : 293.5-296°C

MASS (ESI) (m/z) : 187 (M+H)⁺

¹H-NMR (DMSO-d₆, δ) 4.72(1H,d,J=5.5Hz),
4.77(1H,d,J=5.5Hz), 7.83(1H,dd,J=7.5, 5.5Hz),
8.30(1H,d,J=7.5Hz), 8.37(1H,d,J=7.5Hz), 8.55(2H,br s),
8.81(1H,d,J=7.5Hz), 8.97(1H,s), 9.20(1H,d,J=5.5Hz)

Preparation 43

The object compound was obtained according to a similar manner to that of Preparation 28.

pale yellow solid

MASS (ESI) (m/z) : 434 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.42(9H,s), 3.15(2H,d,J=7.5Hz),
4.50(1H,m), 4.80(1H,dd,J=20.5, 5.5Hz),
4.89(1H,dd,J=20.5, 5.5Hz), 5.03(1H,m), 6.95(1H,m),
7.19-7.35(5H,m), 7.52(1H,dd,J=7.5, 5.5Hz), 8.16-8.27(2H,m),
8.30(1H,d,J=7.5Hz), 8.48(1H,s), 9.07(1H,d,J=5.5Hz)

Preparation 44

The object compound was obtained according to a similar manner to that of Preparation 2.

pale violet amorphous solid

MASS (ESI) (m/z) : 429 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.42(9H,s), 3.05(3H,s),
3.18(1H,dd,J=13.5, 8.5Hz), 3.37(1H,dd,J=13.5, 6.0Hz),
5.03(1H,m), 5.59(1H,br d,J=7.5Hz), 7.03-7.11(2H,m),
7.18(1H,s), 7.20-7.31(3H,m), 7.44(1H,dd,J=7.5, 5.5Hz),
7.57(1H,d,J=7.5Hz), 7.70(1H,s), 8.15(2H,t,J=7.5Hz),
8.95(1H,d,J=5.5Hz)

Preparation 45

The object compound was obtained according to a similar manner to that of Preparation 3.

pale yellow oil

MASS (ESI) (m/z) : 329 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.13-3.30(2H,m), 3.27(3H,s),
4.20(1H,t,J=7.5Hz), 7.08-7.15(2H,m), 7.18(1H,s),
7.21-7.34(3H,m) 7.43(1H,dd,J=7.5, 5.5Hz) 7.63(1H,d,J=7.5Hz),
7.73(1H,s), 8.15(2H,t,J=7.5Hz), 8.93(1H,d,J=5.5Hz)

Preparation 46

The object compound was obtained according to a similar manner to that of Preparation 40.

off-white solid

mp : 205-208°C

MASS (ESI) (m/z) : 187 (M+H)⁺

¹H-NMR (CDCl₃-CD₃OD, δ) 2.40(3H,s), 7.59(1H,t,J=7.5Hz),
7.73(1H,t,J=7.5Hz), 7.87(1H,d,J=7.5Hz), 8.10(1H,d,J=7.5Hz),
8.28(1H,d,J=1.0Hz), 9.46(1H,d,J=1.0Hz)

Preparation 47

The object compound was obtained according to a similar manner to that of Preparation 41.

pale brown solid

mp : 165-174°C

MASS (ESI) (m/z) : 341 (M+H)⁺

¹H-NMR (CDCl₃, δ) 2.44(3H,s), 2.47(3H,s), 7.39(1H,d,J=7.5Hz),
7.60(1H,t,J=7.5Hz), 7.79(1H,t,J=7.5Hz), 7.85(1H,d,J=7.5Hz),
7.98(2H,d,J=7.5Hz), 8.11(1H,d,J=7.5Hz), 8.28(1H,d,J=1.5Hz),
9.14(1H,d,J=1.5Hz)

Preparation 48

The object compound was obtained according to a similar manner to that of Preparation 42.

off-white solid

mp : 290-294°C

MASS (ESI) (m/z) : 187 (M+H)⁺

¹H-NMR (DMSO-d₆, δ) 4.75(1H,d,J=5.5Hz), 4.79(1H,d,J=5.5Hz),
7.80(1H,t,J=7.5Hz), 8.02(1H,t,J=7.5Hz), 8.18(1H,d,J=7.5Hz),
8.25(1H,d,J=7.5Hz), 8.61(2H,br s), 9.27(1H,d,J=1.0Hz),
9.41(1H,d,J=1.0Hz)

Preparation 49

The object compound was obtained according to a similar manner to that of Preparation 28.

pale yellow amorphous solid

MASS (ESI) (m/z) : 434 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.43(9H,s), 3.10-3.19(2H,m), 4.51(1H,m),

4.79(1H,dd,J=20.5, 4.5Hz), 4.88(1H,dd,J=20.5, 4.5Hz),
 5.03(1H,m), 6.93(1H,m), 7.17-7.34(5H,m), 7.69(1H,t,J=7.5Hz),
 7.90(1H,t,J=7.5Hz), 7.97(1H,d,J=7.5Hz), 8.18(1H,d,J=7.5Hz),
 8.73(1H,d,J=1.0Hz), 9.40(1H,d,J=1.0Hz)

Preparation 50

The object compound was obtained according to a similar manner to that of Preparation 2.

pale brown amorphous solid

MASS (ESI) (m/z) : 429 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.45(9H,s), 3.03(3H,s),
 3.17(1H,dd,J=13.0, 9.0Hz), 3.39(1H,dd,J=13.0, 5.5Hz),
 5.05(1H,m), 5.63(1H,d,J=7.5Hz), 7.03-7.12(2H,m),
 7.19-7.38(4H,m), 7.60(1H,t,J=7.5Hz), 7.76(1H,t,J=7.5Hz),
 7.83(1H,d,J=7.5Hz), 8.00(1H,d,J=1.0Hz), 8.12(1H,d,J=7.5Hz),
 8.80(1H,d,J=1.0Hz)

Preparation 51

The object compound was obtained according to a similar manner to that of Preparation 3.

pale brown amorphous solid

MASS (ESI) (m/z) : 329 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.18-3.25(2H,m), 3.22(3H,s),
 4.21(1H,t,J=7.5Hz), 7.06-7.13(2H,m), 7.20-7.36(4H,m),
 7.60(1H,t,J=7.5Hz), 7.76(1H,t,J=7.5Hz), 7.83(1H,d,J=7.5Hz),
 8.04(1H,d,J=1.5Hz), 8.12(1H,d,J=7.5Hz), 8.83(1H,d,J=1.5Hz)

Preparation 52

The object compound was obtained according to a similar manner to that of Preparation 1.

mp : 144-146°C

MASS : 413 (M+1)

¹H-NMR (CDCl₃) δ : 1.41(9H,s), 3.00-3.20(2H,m),
 3.87(3H,s), 4.49(1H,d,J=5Hz), 4.53-4.74(2H,m),
 5.08(1H,d,J=5Hz), 6.95(3H,d,J=8Hz), 7.19-7.32(5H,m),

7.92(2H,d,J=8Hz)

Preparation 53

The object compound was obtained according to a similar manner to that of Preparation 2.

mp : 125-128°C

MASS : 408 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ : 1.38(9H,s), 2.93(3H,s),
3.11($1 \times 1/3$ H,d,J=8Hz), 3.17($1 \times 2/3$ H,d,J=8Hz),
3.31($1 \times 2/3$ H,d,J=6Hz), 3.37($1 \times 1/3$ H,d,J=6Hz),
3.83(3H,s), 4.99(1H,q,J=8Hz), 5.59(1H,d,J=8Hz),
6.92(2H,d,J=8Hz), 6.98(1H,s), 7.00-7.10(2H,m)
7.14(2H,d,J=8Hz), 7.20-7.30(3H,m)

Preparation 54

The object compound was obtained according to a similar manner to that of Preparation 3.

oil

MASS : 308 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ : 3.08-3.28(2H,m), 3.12(3H,s),
3.81(3H,s), 4.17(1H,t,J=8Hz), 6.94(2H,d,J=8Hz),
6.99(1H,s), 7.09(2H,d,J=8Hz), 7.11-7.40(5H,m)

Preparation 55

The object compound was obtained according to a similar manner to that of Preparation 32.

colorless solid

mp : 144-150°C

MASS (ESI) (m/z) : 408 (M+H)⁺

$^1\text{H-NMR}$ (CDCl_3 , δ) 1.37(9H,s), 3.20(2H,m), 3.78(3H,s),
4.59(1H,d,J=14.5Hz), 4.70(1H,d,J=14.5Hz), 5.03(1H,m),
5.35(1H,m), 6.61(1H,s), 6.76(2H,d,J=9.0Hz),
6.81(2H,d,J=9.0Hz), 6.97-7.06(3H,m), 7.17-7.23(3H,m)

Preparation 56

The object compound was obtained according to a similar manner to

that of Preparation 3.

off-white oil

MASS (ESI) (m/z) : 308 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.03(1H,dd,J=14.5, 7.5Hz),
3.14(1H,dd,J=14.5, 7.5Hz), 3.77(3H,s), 4.09(1H,t,J=7.5Hz),
4.73(1H,d,J=15.0Hz), 4.81(1H,d,J=15.0Hz), 6.71(1H,s),
6.81(2H,d,J=7.5Hz), 6.91(2H,d,J=7.5Hz), 7.01-7.07(3H,s),
7.19-7.30(3H,m)

Preparation 57

The object compound was obtained according to a similar manner to that of Preparation 28.

pale yellow oil

MASS (ESI) (m/z) : 367 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.47(9H,s), 1.98(1H,m), 2.13(3H,s),
2.16(1H,m), 2.61(2H,t,J=7.5Hz), 4.41(1H,m),
4.77(2H,t,J=4.5Hz), 5.23(1H,m), 7.14(1H,m),
7.50(2H,t,J=7.5Hz), 7.63(1H,t,J=7.5Hz), 7.98(2H,d,J=7.5Hz)

Preparation 58

The object compound was obtained according to a similar manner to that of Preparation 2.

pale brown oil

MASS (ESI) (m/z) : 362 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.43(9H,s), 2.12(3H,s), 2.12-2.61(4H,m),
3.63(3H,s), 5.05-5.26(2H,m), 7.01(1H,s), 7.33-7.51(5H,m)

Preparation 59

The object compound was obtained according to a similar manner to that of Preparation 3.

pale yellow oil

MASS (ESI) (m/z) : 262 (M+H)⁺

¹H-NMR (CDCl₃, δ) 2.08(1H,m), 2.11(3H,s), 2.25(1H,m),
2.55-2.77(2H,m), 3.61(3H,s), 4.20(1H,t,J=7.5Hz), 7.01(1H,s),
7.33-7.48(5H,m)

Preparation 60

To a solution of the starting compound (893 mg) in tetrahydrofuran (4.5 ml) was added 2,4-bis(4-methoxyphenyl)-1,3-dithia-2,4-diphosphetane-2,4-disulfide (552 mg). The mixture was stirred at 50°C for 4.5 hours, then allowed to cool to room temperature and concentrated. The crude product was purified by column chromatography (silica gel, chloroform) to give the object compound as pale orange powder (476 mg).

MASS (ESI) (m/z) : 489, 491 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.41(9H,s), 3.12-3.32(2H,m), 3.76(3H,s), 5.11-5.31(2H,m), 6.80(2H,d,J=8Hz), 7.02(2H,d,J=8Hz), 7.36(2H,d,J=8Hz), 7.50(2H,d,J=8Hz), 7.87(1H,s)

Preparation 61

The object compound was obtained according to a similar manner to that of Preparation 9.

MASS (ESI) (m/z) : 389, 391 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 2.84(1H,dd,J=13 and 9Hz), 3.31(1H,dd,J=13 and 5Hz), 3.78(3H,s), 4.46(1H,dd,J=9 and 5Hz), 6.86(2H,d,J=8Hz), 7.13(2H,d,J=8Hz), 7.40(2H,d,J=8Hz), 7.51(2H,d,J=8Hz), 7.88(1H,s)

Preparation 62

The object compound was obtained according to a similar manner to that of Preparation 28.

mp : 140-143°C

MASS : 427 (M+1)

¹H-NMR (CDCl₃) δ 1.38(9H,s), 1.43(3H,t,J=8Hz), 3.00-3.19(2H,m), 4.11(2H,q,J=8Hz), 4.40-4.72(3H,m), 4.96-5.10(1H,m), 6.90(1H,br s), 6.92(2H,d,J=8Hz), 7.13-7.35(5H,m), 7.91(2H,d,J=8Hz)

Preparation 63

The object compound was obtained according to a similar manner to

that of Preparation 2.

mp : 86-91°C

MASS : 422 (M+1)

¹H-NMR (CDCl₃) δ 1.41(9H,s), 1.42(3H,t,J=8Hz), 2.92(3H,s),
 3.11(1×1/3H,d,J=10Hz), 3.18(1×2/3H,d,J=10Hz),
 3.31(1×2/3H,d,J=6Hz), 3.36(1×1/3H,d,J=6Hz),
 4.05(2H,q,J=8Hz), 5.00(1H,q,J=8Hz), 5.60(1H,d,J=8Hz),
 6.91(2H,d,J=8Hz), 6.99(1H,s), 7.00-7.09(2H,m),
 7.13(2H,d,J=8Hz), 7.19-7.25(3H,m)

Preparation 64

The object compound was obtained according to a similar manner to that of Preparation 3 except that a mixutre of trifluoroacetic acid and dichloromethane was used instead of trifluoroacetic acid.

MASS : 322 (M+1)

¹H-NMR (CDCl₃) δ 1.43(3H,t,J=8Hz), 3.09-3.27(2H,m), 3.12(3H,s),
 4.07(2H,q,J=8Hz), 4.13(1H,t,J=8Hz), 6.91(2H,d,J=8Hz),
 7.00(1H,s), 7.10(2H,d,J=7Hz), 7.19(2H,d,J=8Hz),
 7.21-7.31(3H,m)

Preparation 65

The object compound was obtained according to a similar manner to that of Preparation 28.

amorphous solid

MASS : 411 (M+1)

¹H-NMR (CDCl₃) δ 1.29(3H,t,J=8Hz), 1.40(9H,s),
 2.71(2H,q,J=8Hz), 3.00-3.20(2H,m), 4.40-4.53(1H,m),
 4.58-4.80(2H,m), 5.00-5.15(1H,m), 6.94(1H,s), 7.12-7.40(7H,m),
 7.88(2H,d,J=8Hz)

Preparation 66

The object compound was obtained according to a similar manner to that of Preparation 2.

oil

MASS : 406 (M+1)

¹H-NMR (CDCl₃) δ 1.22(3H,t,J=8Hz), 1.40(9H,s), 2.67(2H,q,J=8Hz), 2.93(3H,s), 3.08-3.20(1H,m), 3.30-3.40(1H,m), 5.00(1H,q,J=8Hz), 5.69(1H,d,J=8Hz), 7.00(1H,s), 7.01-7.10(2H,m), 7.10-7.18(2H,m), 7.18-7.32(5H,m)

Preparation 67

The object compound was obtained according to a similar manner to that of Preparation 64.

oil

MASS : 306 (M+1)

¹H-NMR (CDCl₃) δ 1.30(3H,t,J=8Hz), 2.68(2H,q,J=8Hz), 3.09-3.28(2H,m), 3.18(3H,s), 4.13(1H,t,J=8Hz), 7.01(1H,s), 7.04-7.10(2H,m), 7.12-7.30(7H,m)

Preparation 68

The object compound was obtained according to a similar manner to that of Preparation 28.

oil

MASS : 447 (M+1)

¹H-NMR (CDCl₃) δ 1.40(9H,s), 3.02(2H,d,J=6Hz), 3.76(3H,s), 4.33-4.47(1H,m), 4.50-4.71(2H,m), 4.91-5.30(1H,m), 6.72-6.80(1H,m), 6.81(2H,d,J=8Hz), 7.11(2H,d,J=8Hz), 7.30-7.40(1H,m), 7.41-7.48(2H,m), 7.51(1H,d,J=8Hz)

Preparation 69

The object compound was obtained according to a similar manner to that of Preparation 2.

oil

MASS : 442 (M+1)

¹H-NMR (CDCl₃) δ 1.47(9H,s), 2.86(3H,s), 3.01-3.12(1H,m), 3.22-3.31(1H,m), 3.73(3H,s), 4.89-5.00(1H,m), 5.61(1H,d,J=8Hz), 6.73(2H,d,J=8Hz), 6.97(2H,d,J=8Hz), 7.00(1H,s), 7.20-7.39(3H,m), 7.44(1H,d,J=8Hz)

Preparation 70

The object compound was obtained according to a similar manner to

that of Preparation 64.

oil

MASS : 342 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ 3.04(3H,s), 3.08-3.17(2H,m), 3.75(3H,s),
4.11(1H,t,J=8Hz), 6.80(2H,d,J=8Hz), 7.00(2H,d,J=8Hz),
7.01(1H,s), 7.21-7.40(3H,m), 7.47(1H,d,J=7Hz)

Preparation 71

The object compound was obtained according to a similar manner to that of Preparation 28.

mp : 115-122°C

MASS : 427 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ 1.42(9H,s), 2.42(3H,s), 3.07(2H,d,J=7Hz),
3.76(3H,s), 4.38-4.50(1H,m), 4.58-4.77(2H,m), 4.98-5.10(1H,m),
6.81(2H,d,J=8Hz), 6.87-6.92(1H,m), 7.11(2H,d,J=8Hz),
7.29(2H,d,J=8Hz), 7.85(2H,d,J=8Hz)

Preparation 72

The object compound was obtained according to a similar manner to that of Preparation 2.

oil

MASS : 422 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ 1.42(9H,s), 2.38(3H,s), 2.99(3H,s),
3.01-3.18(1H,m), 3.20-3.30(1H,m), 3.71(3H,s),
4.93(1H,q,J=8Hz), 5.58(1H,d,J=8Hz), 6.73(2H,d,J=8Hz),
6.93(2H,d,J=8Hz), 7.00(1H,s), 7.11(2H,d,J=7Hz),
7.20(2H,d,J=7Hz)

Preparation 73

The object compound was obtained according to a similar manner to that of Preparation 64.

oil

MASS : 322 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ 2.39(3H,s), 3.10(1H,t,J=8Hz), 3.19(3H,s),
3.80(3H,s), 4.12(1H,t,J=8Hz), 6.81(2H,d,J=8Hz),

7.00(2H,d,J=8Hz), 7.01(1H,s), 7.12-7.23(5H,m)

Preparation 74

The object compound was obtained according to a similar manner to that of Preparation 28.

mp : 105-108°C

MASS : 447 (M+1)

¹H-NMR (CDCl₃) δ 1.40(9H,s), 3.06(2H,d,J=7Hz), 3.79(3H,s), 4.41(1H,br s), 4.58-4.77(2H,m), 4.99(1H,br s), 6.81(2H,d,J=8Hz), 6.83(1H,s), 7.12(2H,d,J=8Hz), 7.49(2H,d,J=7Hz), 7.90(2H,d,J=7Hz)

Preparation 75

The object compound was obtained according to a similar manner to that of Preparation 2.

amorphous solid

¹H-NMR (CDCl₃) δ 1.40(9H,s), 2.98-3.13(1H,m), 3.00(3H,s), 3.21-3.32(1H,m), 3.78(3H,s), 4.90-5.02(1H,m), 5.57(1H,d,J=8Hz), 6.78(2H,d,J=8Hz), 6.93(2H,d,J=8Hz), 7.02(1H,s), 7.18(2H,d,J=8Hz), 7.38(2H,d,J=8Hz)

Preparation 76

The object compound was obtained according to a similar manner to that of Preparation 64.

oil

¹H-NMR (CDCl₃) δ 3.11(2H,t,J=7Hz), 3.19(3H,s), 3.80(3H,s), 4.11(1H,t,J=8Hz), 6.80(2H,d,J=8Hz), 7.00(2H,d,J=8Hz), 7.02(1H,s), 7.20(2H,d,J=8Hz), 7.38(2H,d,J=8Hz)

Preparation 77

The object compound was obtained according to a similar manner to that of Preparation 28.

amorphous solid

MASS : 447 (M+1)

¹H-NMR (CDCl₃) δ 1.40(9H,s), 3.07(2H,d,J=6Hz), 3.73(3H,s), 4.42(1H,br s), 4.58-4.80(2H,m), 5.01(1H,br s),

6.81(2H,d,J=8Hz), 6.84(1H,br s), 7.11(2H,d,J=8Hz),
7.42(1H,t,J=8Hz), 7.59(1H,d,J=8Hz), 7.81(1H,d,J=8Hz),
7.91(1H,s)

Preparation 78

The object compound was obtained according to a similar manner to that of Preparation 2.

amorphous solid

MASS : 442 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ 1.43(9H,s), 3.00(3H,s), 3.11-3.32(2H,m),
3.79(3H,s), 4.91-5.03(1H,m), 5.88(1H,br s), 6.78(2H,d,J=8Hz),
6.93(2H,d,J=8Hz), 7.03-7.19(2H,m), 7.21(1H,s),
7.30-7.40(2H,m)

Preparation 79

The object compound was obtained according to a similar manner to that of Preparation 64.

oil

MASS : 342 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ 3.07-3.20(2H,m), 3.18(3H,s), 3.78(3H,s),
4.20(1H,t,J=8Hz), 6.80(2H,d,J=8Hz), 6.99(2H,d,J=8Hz),
7.09(1H,s), 7.11-7.21(1H,m), 7.28(1H,s), 7.30-7.40(2H,m)

Preparation 80

The object compound was obtained according to a similar manner to that of Preparation 28.

mp : 120-123°C

MASS : 431 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ 1.43(9H,s), 3.08(2H,d,J=8Hz), 3.76(3H,s),
4.42(1H,br s), 4.58-4.78(2H,m), 5.00(1H,br s),
6.82(2H,d,J=8Hz), 6.87(1H,s), 7.10-7.22(4H,m),
8.00(2H,t,J=7Hz)

Preparation 81

The object compound was obtained according to a similar manner to that of Preparation 2.

amorphous solid

MASS : 426 (M+1)

¹H-NMR (CDCl₃) δ 1.41(9H,s), 2.99(3H,s), 3.01-3.32(2H,m),
3.74(3H,s), 4.90-5.02(1H,m), 5.70(1H,d,J=7Hz),
6.76(2H,d,J=8Hz), 6.95(2H,d,J=8Hz), 7.01(1H,s),
7.03-7.16(2H,m), 7.16-7.23(2H,m)

Preparation 82

The object compound was obtained according to a similar manner to that of Preparation 64.

oil

MASS : 326 (M+1)

¹H-NMR (CDCl₃) δ 3.08-3.22(2H,m), 3.18(3H,s), 3.80(3H,s),
4.18(1H,t,J=8Hz), 6.80(2H,d,J=8Hz), 6.99(2H,d,J=8Hz),
7.00(1H,s), 7.09(2H,t,J=8Hz), 7.20-7.30(2H,m)

Preparation 83

The object compound was obtained according to a similar manner to that of Preparation 28.

mp : 131-134°C

MASS : 457 (M+1)

¹H-NMR (CDCl₃) δ 1.43(9H,s), 1.47(3H,t,J=8Hz), 3.05(2H,d,J=8Hz),
3.77(3H,s), 4.10(2H,q,J=8Hz), 4.41(1H,br s), 4.51-4.73(2H,m),
5.01(1H,br s), 6.80(2H,d,J=8Hz), 6.90(1H,br s),
6.92(2H,d,J=8Hz), 7.11(2H,d,J=8Hz), 7.91(2H,d,J=8Hz)

Preparation 84

The object compound was obtained according to a similar manner to that of Preparation 2.

solid

MASS : 452 (M+1)

¹H-NMR (CDCl₃) δ 1.41(9H,s), 1.44(3H,t,J=8Hz), 2.99(3H,s),
3.01-3.13(1H,m), 3.20-3.31(1H,m), 3.78(3H,s),
4.03(2H,q,J=8Hz), 4.88-4.98(1H,m), 5.58(1H,q,J=8Hz),
6.78(2H,d,J=8Hz), 6.88-7.00(5H,m), 7.12(2H,d,J=8Hz)

Preparation 85

The object compound was obtained according to a similar manner to that of Preparation 64.

oil

MASS : 352 (M+1)

¹H-NMR (CDCl₃) δ 1.43(3H,t,J=8Hz), 3.02-3.17(2H,m), 3.18(3H,s), 3.75(3H,s), 4.00-4.18(1H,m), 4.05(2H,q,J=8Hz), 6.80(2H,d,J=8Hz), 6.91(2H,d,J=8Hz), 6.98(1H,s), 7.00(2H,d,J=8Hz), 7.19(2H,d,J=8Hz)

Preparation 86

A solution of potassium tert-butoxide (4.2 g) in anhydrous tetrahydrofuran (70 ml) was cooled under nitrogen atmosphere to -70°C, and a solution of the starting compound (10 g) in anhydrous tetrahydrofuran (35 ml) was added while maintaining the reaction temperature at -70°C. After 30 minutes, this solution was added dropwise to a solution of 4-bromobenzoyl chloride (8.21 g) in anhydrous tetrahydrofuran (24 ml) with stirring while cooling at -70°C on a cooling bath. The reaction mixture was stirred at -70°C for 1 hour and quenched with 3N-hydrochloric acid (100 ml). The cooling bath was removed and the reaction mixture was concentrated to dryness under reduced pressure. The residue was dissolved in water (15 ml) and extracted with diethyl ether (twice). The aqueous layer was concentrated *in vacuo*, and the residue was dissolved in anhydrous methanol. The precipitated white solid (KCl) was removed by filtration. The filtrate was concentrated *in vacuo* and the residue was crystallized from tetrahydrofuran/diethyl ether to give the object compound as an off-white solid.

mp : 183-188°C

MASS : 286 (M+H)⁺

¹H-NMR (DMSO-d₆, δ) 1.03(3H,t,J=7.0Hz), 4.13(2H,q,J=7.0Hz), 6.24(1H,s), 7.86(2H,d,J=7.5Hz), 8.09(2H,d,J=7.5Hz), 9.10(2H,br s),

Preparation 87

The object compound was obtained according to a similar manner to that of Preparation 28.

pale yellow amorphous solid

MASS : 531 (M-H)⁺

¹H-NMR (CDCl₃, δ) 1.14(3H,t,J=7.0Hz), 1.40(9H,s),
2.97-3.18(2H,m), 4.16(2H,q,J=7.0Hz), 4.49(1H,m), 4.96(1H,m),
6.03(1H×3/7,d,J=7.0Hz), 6.06(1H×4/7,d,J=7.0Hz),
7.14-7.31(6H,m), 7.64(2H,d,J=7.5Hz), 7.95(2H×3/7,d,J=7.5Hz),
7.97(2H×4/7,d,J=7.5Hz)

Preparation 88

The object compound was obtained according to a similar manner to that of Preparation 2.

pale yellow amorphous solid

MASS : 528 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.18(3H,t,J=7.0Hz), 1.41(9H,s), 2.69(3H,s),
3.17(1H,dd,J=13.5 and 9.0Hz), 3.37(1H,dd,J=13.5 and 7.0Hz),
4.23(2H,q,J=7.0Hz), 4.98(1H,m), 5.74(1H,d,J=7.5Hz),
6.97-7.08(4H,m), 7.19-7.27(3H,m), 7.55(2H,d,J=7.5Hz)

Preparation 89

The object compound was obtained according to a similar manner to that of Preparation 3.

pale yellow oil

MASS : 428 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.20(3H,t,J=7.0Hz), 2.97(3H,s),
3.22(2H,d,J=7.0Hz), 4.19(1H,t,J=7.0Hz), 4.25(2H,q,J=7.0Hz),
7.05-7.15(4H,m), 7.21-7.33(3H,m), 7.57(2H,d,J=7.5Hz)

Preparation 90

The object compound was obtained according to a similar manner to that of Preparation 28.

pale yellow solid

mp : 148-152.5°C

MASS : 383 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.41(9H,s), 3.12(2H,d,J=7.0Hz), 4.49(1H,m), 4.65(1H,dd,J=20.5 and 5.5Hz), 4.75(1H,dd,J=20.5 and 5.5Hz), 5.03(1H,m), 6.89(1H,m), 7.28-7.32(5H,m), 7.50(2H,t,J=7.5Hz), 7.62(1H,t,J=7.5Hz), 7.94(2H,d,J=7.5Hz)

Preparation 91

The object compound was obtained according to a similar manner to that of Preparation 2.

brown amorphous solid

MASS : 378 (M+H)⁺

¹H-NMR (CDCl₃, δ) 1.42(9H,s), 2.97(3H,s), 3.14(1H,dd,J=13.5 and 9.0Hz), 3.35(1H,dd,J=13.5 and 7.0Hz), 5.01(1H,m), 5.59(1H,d,J=7.5Hz), 7.01-7.08(2H,m), 7.03(1H,s), 7.17-7.29(5H,m), 7.32-7.44(3H,m)

Preparation 92

The object compound was obtained according to a similar manner to that of Preparation 3.

brown oil

MASS : 278 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.14(1H,dd,J=13.5 and 7.5Hz), 3.18(3H,s), 3.21(1H,dd,J=13.5 and 7.5Hz), 4.15(1H,t,J=7.5Hz), 7.05(1H,s), 7.09(2H,d,J=7.5Hz), 7.19-7.44(8H,m)

Preparation 93

The object compound was obtained according to a similar manner to that of Preparation 28.

MASS (ESI) (m/z) : 503, 505 (M-H)⁻

¹H-NMR (CDCl₃, 300MHz) δ 1.38(3H,t,J=7Hz), 1.41(9H,s), 3.04(2H,d,J=7Hz), 3.98(2H,q,J=7Hz), 4.32-4.49(1H,m), 4.53-4.77(2H,m), 4.99(1H,br d,J=8Hz), 6.80(2H,d,J=8Hz), 6.83(1H,br s), 7.10(2H,d,J=8Hz), 7.62(2H,d,J=8Hz), 7.80(2H,d,J=8Hz)

Preparation 94

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 500, 502 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ 1.39(3H, t, J=7Hz), 1.41(9H, s), 2.99(3H, s), 3.05(1H, dd, J=13 and 9Hz), 3.25(1H, dd, J=13 and 5Hz), 3.98(2H, q, J=7Hz), 4.86-5.02(1H, m), 5.56(1H, br d, J=8Hz), 6.73(2H, d, J=8Hz), 6.91(2H, d, J=8Hz), 7.01(1H, s), 7.09(2H, d, J=8Hz), 7.51(2H, d, J=8Hz)

Preparation 95

The object compound was obtained according to a similar manner to that of Preparation 9.

MASS (ESI) (m/z) : 400, 402 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ 1.40(3H, t, J=7Hz), 3.00-3.18(2H, m), 3.19(3H, s), 4.00(2H, q, J=7Hz), 4.10(1H, t, J=7Hz), 6.80(2H, d, J=8Hz), 6.96(2H, d, J=8Hz), 7.04(1H, s), 7.15(2H, d, J=8Hz), 7.54(2H, d, J=8Hz)

Preparation 96

The object compound was obtained according to a similar manner to that of Preparation 28.

MASS (ESI) (m/z) : 491, 493 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ 1.41(9H, s), 2.92-3.18(2H, m), 3.87(3H, s), 4.40-4.53(1H, m), 4.53-4.78(2H, m), 5.02(1H, br d, J=8Hz), 6.95(2H, d, J=8Hz), 6.98(1H, br s), 7.09(2H, d, J=8Hz), 7.40(2H, d, J=8Hz), 7.93(2H, d, J=8Hz)

Preparation 97

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 486, 488 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ 1.40(9H, s), 3.09(3H, s), 3.10-3.31(2H, m), 3.83(3H, s), 4.91-5.06(1H, m), 5.48(1H, br d, J=8Hz), 6.88-7.01(5H, m), 7.17(2H, d, J=8Hz), 7.35(2H, d, J=8Hz)

Preparation 98

The object compound was obtained according to a similar manner to that of Preparation 9.

MASS (ESI) (m/z) : 386, 388 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ 3.02-3.25(2H,m), 3.23(3H,s), 3.83(3H,s), 4.12(1H,t,J=7Hz), 6.89-7.02(5H,m), 7.20(2H,d,J=8Hz), 7.38(2H,d,J=8Hz)

Preparation 99

The object compound was obtained according to a similar manner to that of Preparation 28.

MASS (ESI) (m/z) : 455 (M-H)⁻

¹H-NMR (CDCl₃, 300MHz) δ 1.39(3H,t,J=7Hz), 1.42(9H,s), 2.96-3.12(2H,m), 3.88(3H,s), 3.98(2H,q,J=7Hz), 4.33-4.51(1H,m), 4.52-4.79(2H,m), 4.93-5.11(1H,m), 6.81(2H,d,J=8Hz), 6.92(1H,br s), 6.95(2H,d,J=8Hz), 7.10(2H,d,J=8Hz), 7.92(2H,d,J=8Hz)

Preparation 100

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 452 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ 1.39(3H,t,J=7Hz), 1.41(9H,s), 2.97(3H,s), 3.00-3.31(2H,m), 3.81(3H,s), 3.98(2H,q,J=7Hz), 4.86-5.01(1H,m), 5.62(1H,br d,J=8Hz), 6.74(2H,d,J=8Hz), 6.85-6.95(4H,m), 6.96(1H,s), 7.15(2H,d,J=8Hz)

Preparation 101

The object compound was obtained according to a similar manner to that of Preparation 9.

MASS (ESI) (m/z) : 352 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ 1.40(3H,t,J=7Hz), 3.00-3.19(2H,m), 3.17(3H,s), 3.82(3H,s), 4.00(2H,q,J=7Hz), 4.10(1H,t,J=7Hz), 6.80(2H,d,J=8Hz), 6.89-7.02(5H,m), 7.20(2H,d,J=8Hz)

Preparation 102

The object compound was obtained according to a similar manner to

that of Preparation 28.

MASS (ESI) (m/z) : 441 (M-H)⁻

¹H-NMR (CDCl₃, 300MHz) δ 1.42(9H,s), 3.06(2H,d,J=7Hz), 3.76(3H,s), 3.88(3H,s), 4.34-4.52(1H,m), 4.54-4.79(2H,m), 4.91-5.10(1H,m), 6.82(2H,d,J=8Hz), 6.91(1H,br s), 6.96(2H,d,J=8Hz), 7.12(2H,d,J=8Hz), 7.93(2H,d,J=8Hz)

Preparation 103

The object compound was obtained according to a similar manner to that of Preparation 2.

MASS (ESI) (m/z) : 438 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ 1.41(9H,s), 2.98(3H,s), 3.01-3.31(2H,m), 3.76(3H,s), 3.81(3H,s), 4.88-5.00(1H,m), 5.59(1H,br d,J=8Hz), 6.77(2H,d,J=8Hz), 6.87-7.00(5H,m), 7.14(2H,d,J=8Hz)

Preparation 104

The object compound was obtained according to a similar manner to that of Preparation 9.

MASS (ESI) (m/z) : 338 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ 3.01-3.20(2H,m), 3.18(3H,s), 3.78(3H,s), 3.83(3H,s), 4.10(1H,t,J=7Hz), 6.81(2H,d,J=8Hz), 6.89-7.05(5H,m), 7.20(2H,d,J=8Hz)

Example 1

To an ice-cooled solution of the starting compound (76 mg), indole-2-carboxylic acid (66 mg) and 1-hydroxybenzotriazole (58 mg) in dichloromethane (1 ml) was added 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (96 mg). The mixture was stirred at room temperature for 12 hours. A saturated aqueous sodium hydrogencarbonate solution was added to the mixture, and then the mixture was extracted three times with chloroform. The organic layer was washed with brine, dried over magnesium sulfate, filtered and concentrated. The residue was purified by column chromatography (silica gel, chloroform/methanol=70/1) to give the object compound as

white powder (128 mg).

MASS (ESI) (m/z) : 331 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.62(3H, s), 4.80(2H, d, J=5Hz),
6.98-7.92(12H, m), 9.50(1H, br s)

Example 2

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 332 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.64(3H, s), 4.80(2H, d, J=5Hz),
7.05(1H, s), 7.20-7.72(12H, m)

Example 3

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 493 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.06(3H, t, J=7Hz), 2.81(3H, s),
3.42-3.65(2H, m), 4.17(2H, q, J=7Hz), 5.48-5.64(1H, m),
6.88-7.63(15H, m), 8.41(1H, br s), 9.50(1H, br s)

Example 4

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 494 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.12(3H, t, J=7Hz), 2.81(3H, s),
3.32-3.56(2H, m), 4.22(2H, q, J=7Hz), 5.48-5.62(1H, m),
7.05-7.70(15H, m), 7.82(1H, br d, J=8Hz)

Example 5

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 451 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.09(3H, s), 3.22-3.50(2H, m),
3.72(3H, s), 5.50-5.64(1H, m), 6.72(2H, d, J=8Hz),
6.96(2H, d, J=8Hz), 7.00-7.65(11H, m), 8.13(1H, br d, J=8Hz),
10.50(1H, br s)

Example 6

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 452 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 3.06(3H,s), 3.17-3.48(2H,m),
3.75(3H,s), 5.41-5.56(1H,m), 6.77(2H,d,J=8Hz),
6.98(2H,d,J=8Hz), 7.10(1H,s), 7.18-7.80(11H,m)

Example 7

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 529, 531 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 3.08(3H,s), 3.22-3.50(2H,m),
3.72(3H,s), 5.50-5.64(1H,m), 6.72(2H,d,J=8Hz),
6.98(2H,d,J=8Hz), 7.00-7.65(10H,m), 8.11(1H,br d,J=8Hz),
9.95(1H,br s)

Example 8

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 530, 532 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 3.06(3H,s), 3.15-3.48(2H,m),
3.75(3H,s), 5.40-5.55(1H,m), 6.77(2H,d,J=8Hz),
6.98(2H,d,J=8Hz), 7.05-7.75(11H,m)

Example 9

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 467 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 2.50(3H,s), 3.01(3H,s),
3.22-3.56(2H,m), 5.51-5.66(1H,m), 6.98-7.68(15H,m),
7.95(1H,br d,J=8Hz), 9.60(1H,br s)

Example 10

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 468 (M+H)⁺
¹H-NMR (CDCl₃, 300MHz) δ : 2.50(3H,s), 3.00(3H,s),
3.22-3.55(2H,m), 5.46-5.60(1H,m), 7.02-7.80(16H,m)

Example 11

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 533, 535 (M+H)⁺
¹H-NMR (CDCl₃, 300MHz) δ : 3.18(3H,s), 3.30-3.48(2H,m),
5.52-5.68(1H,m), 6.93-8.00(15H,m), 9.78(1H,br s)

Example 12

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 534, 536 (M+H)⁺
¹H-NMR (CDCl₃, 300MHz) δ : 3.18(3H,s), 3.26-3.49(2H,m),
5.47-5.61(1H,m), 6.98-7.70(15H,m)

Example 13

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS (ESI) (m/z) : 499 (M+H)⁺
¹H-NMR (CDCl₃+CD₃OD, 300MHz) δ : 2.88(3H,s),
3.00(1×1/3H,d,J=8Hz), 3.03(1×2/3H,d,J=8Hz),
3.11(1×2/3H,d,J=4Hz), 3.16(1×1/3H,d,J=4Hz),
5.30(1H,q,J=6Hz), 6.70-6.90(6H,m), 6.90-7.04(5H,m),
7.10(1H,s), 7.16(1H,d,J=8Hz), 7.26(2H,d,J=8Hz),
7.40(1H,d,J=8Hz)

Example 14

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS (ESI) (m/z) : 500 (M+H)⁺
¹H-NMR (CDCl₃, 300MHz) δ : 2.99(3H,s),

3.30(1×1/3H,d,J=8Hz), 3.32(1×2/3H,d,J=8Hz),
3.49(1×2/3H,d,J=4Hz), 3.51(1×1/3H,d,J=4Hz),
5.49-5.60(1H,m), 7.00-7.19(5H,m), 7.19-7.32(4H,m),
7.40(1H,t,J=8Hz), 7.49(1H,s), 7.52(3H,d,J=8Hz),
7.64(1H,d,J=8Hz), 7.93(1H,d,J=8Hz)

Example 15

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS (ESI) (m/z) : 455 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 3.01(3H,s),
3.32(1×1/3H,d,J=8Hz), 3.39(1×2/3H,d,J=8Hz),
3.49(1×2/3H,d,J=4Hz), 3.52(1×1/3H,d,J=4Hz),
5.60(1H,q,J=8Hz), 7.00-7.19(7H,m), 7.19-7.30(4H,m),
7.30-7.43(3H,m), 7.61(1H,d,J=8Hz), 8.17(1H,d,J=8Hz),
9.88(1H,s)

Example 16

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS (ESI) (m/z) : 456 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 2.99(3H,s),
3.29(1×1/3H,d,J=8Hz), 3.32(1×2/3H,d,J=8Hz),
3.49(1×2/3H,d,J=4Hz), 3.52(1×1/3H,d,J=4Hz),
5.48-5.60(1H,m), 7.03-7.11(3H,m), 7.15(2H,d,J=8Hz),
7.20-7.31(4H,m), 7.38(2H,d,J=8Hz), 7.41-7.58(3H,m),
7.67(1H,d,J=8Hz), 7.80(1H,d,J=8Hz)

Example 17

The object compound was obtained according to a similar manner to that of Example 1.

mp : 145-150°C

MASS (ESI) (m/z) : 435 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 2.31(3H, s), 3.02(3H, s),
3.33-3.57(2H, m), 5.60-5.73(1H, m), 7.00-7.12(7H, m),
7.12-7.22(6H, m), 7.36(1H, d, J=8Hz), 7.59(1H, d, J=8Hz),
8.57(1H, d, J=8Hz)

Example 18

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS (ESI) (m/z) : 436 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 2.38(3H, s), 3.00(3H, s),
3.30(1×1/3H, d, J=8Hz), 3.38(1×2/3H, d, J=8Hz),
3.50(1×2/3H, d, J=4Hz), 3.52(1×1/3H, d, J=4Hz),
5.48-5.62(1H, m), 7.02-7.14(5H, m), 7.16-7.33(6H, m),
7.35-7.55(3H, m), 7.65(1H, d, J=8Hz), 7.91(1H, d, J=8Hz)

Example 19

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS (ESI) (m/z) : 455 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.18(3H, s), 3.40-3.50(2H, m),
5.70(1H, q, J=8Hz), 6.98-7.29(10H, m), 7.30-7.42(4H, m),
7.59(1H, d, J=8Hz), 8.60(1H, d, J=8Hz)

Example 20

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS (ESI) (m/z) : 456 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.19(3H, s), 3.30-3.51(2H, m),
5.49-5.60(1H, m), 7.04(2H, d, J=8Hz), 7.10(1H, s),
7.14-7.31(5H, m), 7.31-7.52(6H, m), 7.64(1H, d, J=8Hz),
7.78(1H, d, J=8Hz)

Example 21

The object compound was obtained according to a similar manner to that of Example 1.

colorless solid

mp : 223-226°C

MASS (ESI) (m/z) : 435 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 2.23(3H,s), 3.23-3.40(2H,m),
4.77(1H,d,J=16.0Hz), 4.83(1H,d,J=16.0Hz), 5.60(1H,q,J=7.5Hz),
6.70(1H,s), 6.78(2H,d,J=7.5Hz), 6.93(1H,s), 6.97-7.29(10H,m),
7.37(1H,d,J=7.5Hz), 7.58(1H,d,J=7.5Hz), 7.62(1H,d,J=7.5Hz),
9.47(1H,br s)

Example 22

The object compound was obtained according to a similar manner to that of Example 1.

pale yellow amorphous solid

MASS (ESI) (m/z) : 421 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.00(3H,s), 3.30(1H,dd,J=12.0, 8.5Hz),
3.49(1H,dd,J=12.0, 5.5Hz), 5.57(1H,m), 6.99-7.43(15H,m),
7.63(1H,d,J=7.5Hz), 7.76(1H,d,J=7.5Hz), 9.41(1H,s)

Example 23

The object compound was obtained according to a similar manner to that of Example 1.

colorless solid

mp : 234-239°C

MASS (ESI) (m/z) : 345 (M+H)⁺

¹H-NMR (CDCl₃-CD₃OD, δ) 3.17(3H,s), 3.20(1H,dd,J=13.5, 9.0Hz),
3.34(1H,dd,J=13.5, 5.5Hz), 5.49(1H,dd,J=9.0, 5.5Hz),
6.66(1H,s), 6.97-7.03(3H,m), 7.13(1H,t,J=7.5Hz),
7.18-7.31(5H,m), 7.41(1H,d,J=7.5Hz), 7.68(1H,d,J=7.5Hz)

Example 24

The object compound was obtained according to a similar manner to that of Example 1.

colorless solid

mp : 251-256°C

MASS (ESI) (m/z) : 331 (M+H)⁺

¹H-NMR (CDCl₃-CD₃OD, δ) 3.31(2H,d,J=7.5Hz), 5.39(1H,t,J=7.5Hz), 6.90(2H,s), 7.02-7.31(8H,m), 7.39(1H,d,J=7.5Hz), 7.64(1H,d,J=7.5Hz)

Example 25

The object compound was obtained according to a similar manner to that of Example 1.

off-white solid

mp : 202-206°C

MASS (ESI) (m/z) : 472 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.10(3H,s), 3.35(1H,dd,J=13.5, 8.5Hz), 3.53(1H,dd,J=13.5, 5.5Hz), 5.61(1H,m), 7.03(1H,s), 7.09-7.17(3H,m), 7.20(1H,s), 7.23-7.32(4H,m), 7.38-7.46(2H,m), 7.56(1H,dd,J=7.5, 2.5Hz), 7.65(1H,d,J=7.5Hz), 7.67(1H,s), 7.75(1H,d,J=7.5Hz), 8.11(2H,d,J=7.5Hz), 8.93(1H,d,J=5.5Hz), 9.40(1H,s)

Example 26

The object compound was obtained according to a similar manner to that of Example 1.

off-white amorphous solid

MASS (ESI) (m/z) : 472 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.07(3H,s), 3.33(1H,dd,J=13.5, 10.0Hz), 3.55(1H,dd,J=13.5, 5.5Hz), 5.62(1H,m), 7.03(1H,s), 7.07-7.18(3H,m), 7.22-7.33(5H,m), 7.41(1H,d,J=7.5Hz), 7.60(1H,t,J=7.5Hz), 7.69(2H,t,J=7.5Hz), 7.77(1H,t,J=7.5Hz), 7.82(1H,d,J=7.5Hz), 8.02(1H,d,J=1.0Hz), 8.13(1H,d,J=7.5Hz), 8.80(1H,d,J=1.0Hz), 9.37(1H,br s)

Example 27

The object compound was obtained according to a similar manner to that of Example 1.

pale yellow amorphous solid

MASS (ESI) (m/z) : 451 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.08(3H,s), 3.38(1H,dd,J=13.5, 9.0Hz),
3.50(1H,dd,J=13.5, 6.0Hz), 3.82(3H,s), 5.64(1H,m),
6.92(2H,d,J=7.5Hz), 7.03-8.14(14H,m), 9.63(1H,br s)

Example 28

The object compound was obtained according to a similar manner to that of Example 1.

colorless solid

mp : 221-230.5°C

MASS (ESI) (m/z) : 451 (M+H)⁺

¹H-NMR (CDCl₃, δ) 3.32(2H,m), 3.70(3H,s), 4.74(2H,s),
5.62(1H,m), 6.67(1H,s), 6.71(2H,d,J=7.5Hz),
6.82(2H,d,J=7.5Hz), 6.93(1H,d,J=1.0Hz), 6.99-7.30(8H,m),
7.37(1H,d,J=7.5Hz), 7.56-7.65(2H,m), 9.50(1H,s)

Example 29

The object compound was obtained according to a similar manner to that of Example 1.

off-white solid

mp : 192.5-198°C

MASS (ESI) (m/z) : 405 (M+H)⁺

¹H-NMR (CDCl₃, δ) 2.10(3H,s), 2.30-2.75(4H,m), 3.66(3H,s),
5.71(1H,q,J=7.5Hz), 6.95-7.04(2H,m), 7.11(1H,t,J=7.5Hz),
7.21-7.47(7H,m), 7.58(1H,d,J=7.5Hz), 7.63(1H,d,J=7.5Hz),
9.54(1H,s)

Example 30

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 532, 534 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.27-3.50(2H,m), 3.74(3H,s),
5.69-5.83(1H,m), 6.79(2H,d,J=8Hz), 6.88(1H,s),
7.04(2H,d,J=8Hz), 7.08-7.69(9H,m), 7.88(1H,s), 9.46(1H,br s)

Example 31

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 533, 535 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.30-3.49(2H,m), 3.75(3H,s),
5.68-5.82(1H,m), 6.79(2H,d,J=8Hz), 7.09(2H,d,J=8Hz),
7.20-7.80(10H,m), 7.89(1H,s)

Example 32

The object compound was obtained according to a similar manner to that of Example 1.

mp : 178-182°C

MASS : 465 (M+1)

¹H-NMR (CDCl₃) δ : 1.42(3H,t,J=8Hz), 3.02(3H,s),
3.36-3.59(2H,m), 4.02(2H,q,J=8Hz), 5.67(1H,q,J=8Hz),
6.89(2H,d,J=8Hz), 7.01(1H,s), 7.03-7.13(6H,m),
7.17-7.30(4H,m), 7.38(1H,d,J=8Hz), 7.60(1H,d,J=8Hz),
8.48(1H,d,J=8Hz)

Example 33

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS : 466 (M+1)

¹H-NMR (CDCl₃) δ : 1.42(3H,t,J=8Hz), 2.95(3H,s),
3.23-3.37(1H,m), 3.43-3.53(1H,m), 4.02(2H,q,J=8Hz),
5.45-5.58(1H,m), 6.90(2H,d,J=8Hz), 7.01(1H,s),
7.03-7.18(4H,m), 7.19-7.31(4H,m), 7.40(1H,t,J=8Hz),
7.43(1H,s), 7.51(1H,d,J=8Hz), 7.63(1H,d,J=8Hz),
7.81(1H,d,J=8Hz)

Example 34

The object compound was obtained according to a similar manner to that of Example 1.

mp : 174-178°C

MASS : 449 (M+1)

¹H-NMR (CDCl₃) δ : 1.28(3H,t,J=8Hz), 2.69(2H,q,J=8Hz),
3.08(3H,s), 3.40-3.60(2H,m), 5.68-5.80(1H,m), 7.02-7.19(7H,m),
7.19-7.30(6H,m), 7.40(1H,d,J=8Hz), 7.61(1H,d,J=8Hz),
8.69(1H,d,J=8Hz)

Example 35

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS : 450 (M+1)

¹H-NMR (CDCl₃) δ : 1.24(3H,t,J=8Hz), 2.69(2H,t,J=8Hz),
3.00(3H,s), 3.25-3.38(1H,m), 3.43-3.57(1H,m), 5.48-5.60(1H,m),
7.00-7.19(5H,m), 7.19-7.32(6H,m), 7.40(1H,t,J=8Hz),
7.45(1H,s), 7.51(1H,d,J=8Hz), 7.63(1H,d,J=8Hz),
7.81(1H,d,J=8Hz)

Example 36

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS : 485 (M+1)

¹H-NMR (CDCl₃) δ : 2.93(3H,s), 3.30-3.50(2H,m), 3.70(3H,s),
5.53-5.63(1H,m), 6.71(2H,d,J=8Hz), 6.98(2H,d,J=8Hz),
7.00-7.12(3H,m), 7.16-7.40(5H,m), 7.42(1H,d,J=8Hz),
7.60(1H,d,J=8Hz), 8.40(1H,d,J=8Hz)

Example 37

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS : 465 (M+1)

¹H-NMR (CDCl₃) δ : 2.39(3H,s), 3.10(3H,s), 3.30-3.50(2H,m),
3.70(3H,s), 5.61(1H,q,J=8Hz), 6.70(2H,d,J=8Hz),
6.99(2H,d,J=8Hz), 7.01-7.28(8H,m), 7.38(1H,d,J=8Hz),
7.60(1H,d,J=8Hz), 8.42(1H,d,J=8Hz)

Example 38

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS : 485 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ : 3.09(3H,s), 3.30-3.50(2H,m), 3.70(3H,s),
5.62(1H,q,J=8Hz), 6.70(2H,d,J=8Hz), 6.99(2H,d,J=8Hz),
7.01-7.29(6H,m), 7.29-7.40(3H,m), 7.59(1H,d,J=8Hz),
8.51(1H,d,J=8Hz)

Example 39

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS : 485 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ : 3.10(3H,s), 3.31-3.52(2H,m), 3.70(3H,s),
5.60-5.72(1H,m), 6.73(2H,d,J=8Hz), 7.01(2H,d,J=8Hz),
7.07-7.20(4H,m), 7.20-7.30(2H,m), 7.30-7.50(3H,m),
7.61(1H,d,J=8Hz), 8.59(1H,d,J=8Hz)

Example 40

The object compound was obtained according to a similar manner to that of Example 1.

amorphous solid

MASS : 469 (M+1)

$^1\text{H-NMR}$ (CDCl_3) δ : 3.08(3H,s), 3.30-3.40(2H,m), 3.71(3H,s),
5.67(1H,q,J=8Hz), 6.71(2H,d,J=8Hz), 7.00(2H,d,J=8Hz),
7.03-7.30(8H,m), 7.39(1H,d,J=8Hz), 7.60(1H,d,J=8Hz),
8.60(1H,d,J=8Hz)

Example 41

The object compound was obtained according to a similar manner to that of Example 1.

mp : 115-118°C

MASS : 495 (M+1)

¹H-NMR (CDCl₃) δ : 1.42(3H,t,J=8Hz), 3.03(3H,s),
 3.20-3.31(1H,m), 3.36-3.47(1H,m), 3.70(3H,s),
 4.03(2H,q,J=8Hz), 5.48-5.59(1H,m), 6.73(2H,d,J=8Hz),
 6.90(2H,d,J=8Hz), 6.99(2H,d,J=8Hz), 7.00(2H,s),
 7.08-7.18(3H,m), 7.23(1H,t,J=8Hz), 7.39(1H,d,J=8Hz),
 7.61(1H,d,J=8Hz), 7.86(1H,d,J=8Hz), 9.60(1H,s)

Example 42

The object compound was obtained according to a similar manner to that of Example 1.

mp : >250°C

MASS : 529 (M+1)

¹H-NMR (CDCl₃) δ : 3.17-3.40(2H,m), 3.52(3H,s), 3.68(3H,s),
 5.49(1H,q,J=8Hz), 6.79(2H,d,J=8Hz), 7.01-7.18(2H,m),
 7.07(1H,s), 7.21(2H,d,J=8Hz), 7.36(2H,d,J=8Hz),
 7.39(1H,t,J=8Hz), 7.61(2H,d,J=8Hz), 8.09(1H,d,J=8Hz),
 8.19(1H,d,J=8Hz), 8.39(1H,d,J=8Hz)

Example 43

The object compound was obtained according to a similar manner to that of Example 1.

pale yellow amorphous solid

MASS : 571 (M+H)⁺

¹H-NMR (CDCl₃) δ : 1.16(3H,t,J=7.0Hz), 2.79(3H,s),
 3.42(1H,dd,J=12.0 and 10.0Hz), 3.53(1H,dd,J=12.0 and 5.5Hz),
 4.22(2H,q,J=7.0Hz), 5.53(1H,m), 6.98(1H,d,J=1.0Hz),
 7.04-7.10(4H,m), 7.11(1H,t,J=7.5Hz), 7.20-7.30(4H,m),
 7.33(1H,d,J=7.5Hz), 7.56(2H,d,J=7.5Hz), 7.64(1H,d,J=7.5Hz),
 7.91(1H,br d,J=7.5Hz), 9.21(1H,br s)

Example 44

The object compound was obtained according to a similar manner to that of Example 1.

off-white solid

mp : 258.5-260°C

MASS : 421 (M+H)⁺

¹H-NMR (CDCl₃) δ : 3.02(3H,s), 3.29(1H,dd,J=13.0 and 8.5Hz),
3.49(1H,dd,J=13.0 and 5.5Hz), 5.58(1H,m), 7.02-7.09(3H,m),
7.10(1H,s), 7.15(1H,d,J=7.5Hz), 7.20-7.43(10H,m),
7.66(1H,d,J=7.5Hz), 7.73(1H,d,J=7.5Hz), 9.48(1H,s)

Example 45

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 543, 545 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 1.39(3H,t,J=7Hz), 3.06(3H,s),
3.25(1H,dd,J=13 and 9Hz), 3.41(1H,dd,J=13 and 5Hz),
3.97(2H,q,J=7Hz), 5.46-5.61(1H,m), 6.75(2H,d,J=8Hz),
6.95(2H,d,J=8Hz), 7.00-7.70(10H,m), 7.90(1H,br d,J=8Hz),
9.55(1H,br s)

Example 46

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 544, 546 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 1.40(3H,t,J=7Hz), 3.04(3H,s),
3.22(1H,dd,J=13 and 9Hz), 3.41(1H,dd,J=13 and 5Hz),
3.98(2H,q,J=7Hz), 5.41-5.55(1H,m), 6.77(2H,d,J=8Hz),
6.98(2H,d,J=8Hz), 7.05-7.75(11H,m)

Example 47

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 529, 531 (M+H)⁺

¹H-NMR (CDCl₃,300MHz) δ : 3.15(3H,s), 3.29-3.48(2H,m),
3.81(3H,s), 5.52-5.66(1H,m), 6.91(2H,d,J=8Hz),
6.97(2H,d,J=8Hz), 7.00(1H,s), 7.02-7.68(9H,m),
8.01(1H,br d,J=8Hz), 9.84(1H,br s)

Example 48

The object compound was obtained according to a similar manner to

that of Example 1.

MASS (ESI) (m/z) : 530, 532 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.12(3H, s), 3.25-3.48(2H, m),
3.82(3H, s), 5.45-5.60(1H, m), 6.93(2H, d, J=8Hz),
6.99(2H, d, J=8Hz), 7.03(1H, s), 7.11-7.70(10H, m)

Example 49

The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 495 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 1.39(3H, t, J=7Hz), 3.02(3H, s),
3.18-3.48(2H, m), 3.82(3H, s), 3.96(2H, q, J=7Hz),
5.45-5.59(1H, m), 6.74(2H, d, J=8Hz), 6.91(2H, d, J=8Hz),
6.95(2H, d, J=8Hz), 7.01(1H, s), 7.02-7.68(7H, m),
7.88(1H, br d, J=8Hz), 9.59(1H, br s)

Example 50

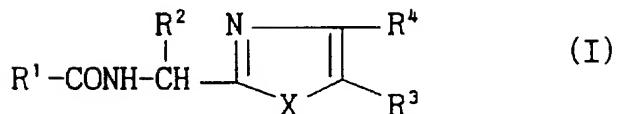
The object compound was obtained according to a similar manner to that of Example 1.

MASS (ESI) (m/z) : 481 (M+H)⁺

¹H-NMR (CDCl₃, 300MHz) δ : 3.04(3H, s), 3.19-3.48(2H, m),
3.74(3H, s), 3.82(3H, s), 5.47-5.61(1H, m), 6.74(2H, d, J=8Hz),
6.91(2H, d, J=8Hz), 6.98(2H, d, J=8Hz), 7.01(1H, s),
7.02-7.68(7H, m), 7.92(1H, br d, J=8Hz), 9.66(1H, br s)

CLAIMS

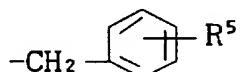
1. A compound of the formula :



wherein

R' is indolyl or benzofuranyl;

R^2 is hydrogen, lower alkylthio(lower)alkyl or a group of the formula:



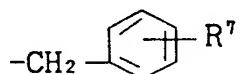
in which R⁵ is hydrogen, lower alkoxy or halogen;

R^3 is hydrogen, quinolyl or phenyl which may have a suitable substituent selected from the group consisting of lower alkyl, lower alkoxy, lower alkylthio and halogen;

R^4 is hydrogen or optionally esterified carboxy; and

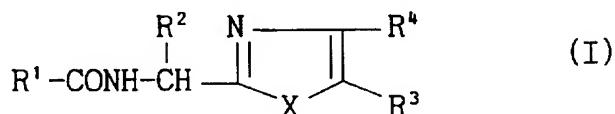
X is S or NR⁶

in which R⁶ is hydrogen, lower alkyl or a group of the formula:



in which R⁷ is lower alkyl or lower alkoxy, and a pharmaceutically acceptable salt thereof.

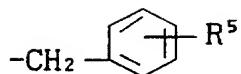
2. A process for preparing a compound of the formula:



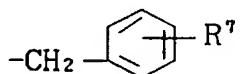
wherein

R' is indolyl or benzofuranyl;

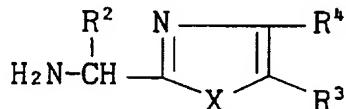
R^2 is hydrogen, lower alkylthio(lower)alkyl or a group of the formula:



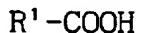
in which R⁵ is hydrogen, lower alkoxy or halogen;
 R³ is hydrogen, quinolyl or phenyl which may have a suitable
 substituent selected from the group consisting of lower alkyl,
 lower alkoxy, lower alkylthio and halogen;
 R⁴ is hydrogen or optionally esterified carboxy; and
 X is S or NR⁶
 in which R⁶ is hydrogen, lower alkyl or a group of the formula:



in which R⁷ is lower alkyl or lower alkoxy,
 or a salt thereof,
 which comprises reacting a compound of the formula:



wherein R², R³, R⁴ and X are each as defined above, or its reactive derivative, or a salt thereof, with a compound of the formula:



wherein R¹ is as defined above, or its reactive derivative, or a salt thereof.

3. A pharmaceutical composition comprising the compound of Claim 1 or a pharmaceutically acceptable salt thereof in admixture with a pharmaceutically acceptable carrier.

4. Use of the compound of Claim 1 or a pharmaceutically acceptable salt thereof as a medicament.

5. Use of the compound of Claim 1 or a pharmaceutically acceptable salt thereof as a medicament for prophylactic and therapeutic treatment of NO-mediated diseases.

INTERNATIONAL SEARCH REPORT

Internat'l Application No
PCT/JP 97/01757

A. CLASSIFICATION OF SUBJECT MATTER					
IPC 6	C07D403/12	A61K31/415	A61K31/425	C07D401/14	C07D405/12
	C07D417/12				

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 C07D C07K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 96 01817 A (ASTRA AB ;MACDONALD JAMES EDWIN (US); SHAKESPEARE WILLIAM CALVIN () 25 January 1996 see page 43; claim 1 see page 46; claims 15-17 ---	1-5
P, Y	WO 96 16981 A (FUJISAWA PHARMACEUTICAL CO ;ITOH YOSHIKUNI (JP); IWAMOTO TOSHIRO () 6 June 1996 see page 689 - page 692; claim 1 ---	1-5
Y	TETRAHEDRON LETTERS, vol. 34, no. 12, 19 March 1993, OXFORD GB, pages 1901-1904, XP002038851 T.D. GORDON ET AL.: "Synthetic Approaches to the 'Azole' Peptide Mimetics" see page 1901, paragraph 1 -----	1-5

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

1

Date of the actual completion of the international search

26 August 1997

Date of mailing of the international search report

09.09.97

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Authorized officer

Fink, D

INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP 97/01757

Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. Claims Nos.: because they relate to subject matter not required to be searched by this Authority, namely:
Remark: Although claim(s) 4 and 5 is(are) directed to a method of treatment of the human/animal body, the search has been carried out and based on the alleged effects of the compound/composition.
2. Claims Nos.: because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:
3. Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
2. As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
4. No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest The additional search fees were accompanied by the applicant's protest. No protest accompanied the payment of additional search fees.

INTERNATIONAL SEARCH REPORT

Information on patent family members

Internal Application No

PCT/JP 97/01757

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9601817 A	25-01-96	AU 2413995 A EP 0759027 A FI 964463 A NO 964698 A	09-02-96 26-02-97 06-11-96 06-11-96
WO 9616981 A	06-06-96	AU 3993795 A ZA 9510201 A	19-06-96 25-06-96